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Tuning the structure and electro-optical properties of α-Cr₂O₃ films by heat treatment/La doping for optoelectronic applications

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Abstract: Controlling the physicochemical features of chromium sesquioxide α -Cr₂O₃ (chromia) films is essential to widening their utilization. This work investigates the influence of heat treatment (HT) and La doping on the physical characteristics of chromia films spin-deposited on glass. X-ray diffraction/Fourier transform infrared spectroscopy and field-emission transmission electron microscopy techniques reveal that the crystallinity, size, and modes of vibration in chromia (eskolaite phase) can be tuned by HT and La doping. Energy-dispersive X-ray analysis confirms the presence of La in the hexagonal chromia films. All films display high transmittance (T%) and improved absorption indices in the UV and near-infrared regions. The films' reflection and refractive indices are in the range of 6-15% and 1.77-1.97, respectively. The impact of HT and La content on the optical band gap, lattice dielectric constant, and electron-effective mass is reported. The currentvoltage characteristic curves reveal the Ohmic resistance, and the films' sheet resistance demonstrates sensitivity to HT and La content. The findings of this study illustrate the possible development of optoelectronic devices and chromia-based IR sensors.

Keywords: α -Cr₂O₃, band gap engineering, optical constants, La doping, I–V characteristics

1 Introduction

Various cutting-edge technologies, such as green energy, energy storage, networks, and superconductivity, have some form of metal oxide coatings or thin films [1]. Metal oxides are fascinating multifunctional materials that could be introduced in various technological applications. The filled s-shells acquire good chemical and thermal stability, and their partially filled *d*-shells provide interesting electronic properties [2]. Among the metal oxides, chromium sesquioxide (α -Cr₂O₃ or chromia) is an attractive multifunctional material due to its low cost (wide availability), low humidity dependency, good chemical and color stability, high thermal stability and corrosion resistance, and adjustable band structure [3]. Chromia is an antiferromagnetic insulator with a Neel temperature of 307 K (34°C) [4]. These properties make chromia suitable for several applications, such as a hard coating or protection layer for tools and machines for high-temperature applications [5]. According to Tian et al. [6], adding 0.1% chromia to ZnO-based varistor ceramics improves I-V nonlinearity and reduces the leakage current. Owing to its interesting photoelectric properties, Li et al. [7] suggested the use of microfibers decorated with chromia, grown by magnetron sputtering, as a saturable absorber for ultrafast fiber laser technology. Alotabi et al. [8] used the photo-deposited chromia layer on different material particles to prevent the reverse reaction in photocatalytic water splitting.

A fundamental step in the application of chromia coatings is controlling the synthesized material by adjusting the processing technique [9]. In other words, the preparation technique and preparative parameters determine the physicochemical and biological features of chromia nanoparticles (NPs) and thin films, and introducing foreign atoms (dopants) can enhance and broaden these features. Heating $Cr(NO_3)_3$ at above 350°C in KOH solution resulted in green nano-crystalline Cr_2O_3 pigments for reflection purposes in the wavelength region 750–2,500 nm [10]. Tsegay *et al.* [11] utilized drop-casting and spin-coating to obtain

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quasi-spherical, nanorod, and meso-spherical-shaped chromia that can act as a spectrally selective solar absorber. The pulsedlaser deposited (PLD) Ni-doped chromia films showed improved photoresponse for solar cells and photoelectrodes [12]. According to Erickson et al. [13], B doping and adjusting the Cr_2O_3/V_2O_3 film thickness, via PLD, can be used to increase the Neel temperature of Cr₂O₃ for voltage control due to its antiferromagnetic properties. The nano-sized Fe-doped chromia exhibited weak ferromagnetism and therefore can be used in logic and spintronic data storage devices [14]. In addition, Bhardwaj et al. [15,16] prepared chromia NPs by the co-precipitation method and found that doping with 10% Co and Mn can reduce the E_{g} value from 2.96 to 2.82 eV, or 2.51 eV, which made the doping materials useful for optoelectronic applications. Moreover, the PLD films made from these materials demonstrated p-type conductivity and room-temperature (RT) ferromagnetism, making them suitable for spintronic device applications. By covering chromia NPs made by the microemulsion method with Ni or Co NPs, Yıldırım [17] improved their specific capacitance and ability to store energy. The X-ray diffraction (XRD) peak intensities of Cr₂O₃ in the composites were greatly reduced compared with those of the uncovered Cr₂O₃ NPs. Furthermore, Gyawali et al. [3] prepared Ag/ chromia NPs by liquid impregnation and annealing with enhanced antibacterial efficiency. Additionally, the Pd/ chromia electrode prepared by reactive dc magnetron cosputtering was found to be a viable candidate for utilization in supercapacitors as future energy storage devices [18].

According to Zekaik *et al.* [19], the $E_{\rm g}$ of 0.0–12 at% Cu-doped chromia in the dip-coated films reduced from ~2.9 to 2.5 eV and the resistance from 87 to 12.8 Ω . Ishtiaq *et al.* [20] found that the hardness of the chromia films deposited on soda lime glass substrates improved with Cu doping to 2.0%, and the antibacterial activity of the films increased with increasing Cu/Cr ratio. Bio-synthesized chromia NPs, 15–24 nm in size, showed $E_{\rm g}$ values in the range of 3.02–3.29 eV [21]. Furthermore, Ba inclusion was found to improve the rate of H₂ production of chromia from NaBH₄ methanolysis from 5,984 to 31,176 mL/g min at 15% Ba content [22].

The rare earth (RE) ions can form strong bonds with the metal oxide's functional groups due to the presence of 4*f* empty orbitals [23], and therefore RE dopants can alter the physical characteristics of chromia. A few reports on the effect of RE on the chromia films' features are recounted in the literature. Ikram *et al.* [23] studied the impact of Sr doping on the bactericidal potential and catalytic activity of nano-sized chromia for wastewater treatment. Jia *et al.* [5] used Y-doping and the ion-plating technique to fabricate chromia films with improved friction and wear resistance performance, as well as reasonable stability at high temperatures. The trivalent doping could induce vacancy complexes (vacant Cr sites proximal to oxygen vacancies), and these structural defects may encourage the researchers to develop spintronic devices based on Cr_2O_3 doped with La^{3+} [24]. In addition, the La electronic 4f-4f transitions result in strong emission peaks with wavelengths in the visible and infrared regions. This enhances the photoluminescence features of the host material for optoelectronic applications [25].

The current work aims to investigate the influence of HT and La doping on the structure and electrical and optical properties of chromia thin films, which have not been reported yet. Most of the physical and chemical synthetic techniques require specialized devices with high costs and high temperatures. However, the sol-gel preparation combined with the spin-coating technique is more affordable and cost-efficient [26]. The solutions were sol-gel-prepared and then spin-coated on glass substrates. The influence of HT/La doping on the structural and morphological features of the films was studied by XRD/Fourier transform infrared (FTIR) spectroscopy and field-emission scanning electron microscopy (FESEM) and energy-dispersive X-ray spectroscopy (EDAX/EDS). The results of optical and electrical properties revealed the possible development of these films for photoelectrooptical devices.

2 Experimental section

2.1 Film preparation

Chromium acetate ($Cr_2C_8H_{16}O_{10}$, ~376 g mol⁻¹, >97%; LOBA Chemie, India), 2-methoxyethanol (C₃H₈O₂, ~76 g/mol; Merck), and monoethanolamine (C₂H₇NO, ~61 g/mol; Merck) were used as the precursor, solvent, and stabilizer, respectively, for chromia thin film preparation. LaCl₃.7H₂O (371.4 g/mol, >97%; WINLAB, India) was used for La doping. About 0.94 g of Cr source was dissolved in 10 mL of solvent (0.25 M) using a magnetic stirrer. Some drops of stabilizing agent were added during stirring for 1.0 h at 45°C until a clear and homogeneous solution was formed. To prepare 2.5 and 5.0 at% La-doped chromia films, the same procedure was repeated by adding the required mass of the La source. The solutions were kept at RT for 1 day. The glass substrates were washed with neat ethanol and neat water, 5 min for each, using an ultrasonic bath and then air dried. The prepared solutions were coated at 2,100 rpm for 25 s. The deposited (six) layers were preheated at 220°C on a hotplate for 5 min to yield films with

200-230 nm thickness. Finally, the samples were heat-treated at 400, 500, and 600°C for 2 h in a furnace. The La-doped films were also annealed at 600°C. The furnace was left to cool naturally to RT. In the following sections, the names (codes) of the samples will be Cr/400, Cr/500, Cr/600, 2.5% La-Cr/600, and 5.0% La-Cr/600.

2.2 Devices

An X'Pert Pro (PANalytical) XRD system was used to examine the phase structure, purity, and size of the spindeposited chromia films. The surface morphology and elemental (chemical) composition were investigated using a ZEISS SUPRA 55-VP field-emission scanning electron microscope attached to an EDS unit. The chromia functional groups and vibration modes were studied by gathering the FTIR spectra using a VERTEX 70 Bruker spectrophotometer in the range of 400–4,000 cm⁻¹. The optical reflectance/transmittance (T%) in the 200–1.500 nm wavelength region was obtained using a UV-3600 Shimadzu spectrophotometer, in normal incidence mode. The reflectance spectrum was collected using an integrating sphere attachment, and a barium compound was used as reference for 100% reflectance. A Keithley-2400 unit was used for obtaining the current (I)-voltage (V) curves using the two-point probe method.

3 Results and discussion

3.1 Structural analysis (XRD, FTIR, and FE-SEM)

The structural properties of the coated chromia films that were annealed at 400-600°C and doped with La were investigated using the X-ray diffractometer in the 2θ range of 20–75° and applying a Cu K_{α} radiation source (wavelength: $\lambda \sim 0.1541$ nm). The obtained diffraction patterns are shown in Figure 1. The peak intensity grew significantly with the heat treatment (HT), indicating the improvement in crystallinity with HT at 400-600°C. The most significant peaks were observed at $2\theta = 24.66, 33.57$, 36.21, and 54.83°. However, doping with La reduced the peak intensity again, indicating deterioration of crystallinity; however, the Cr₂O₃ phase remained unaffected with doping. Similar patterns were obtained for undoped and Cu-doped chromia films prepared by spray pyrolysis on



Figure 1: XRD patterns of chromia films annealed at 400, 500, and 600°C and doped with 2.5 and 5.0 at% La.

pre-heated (400°C) soda lime glass substrates [20]. In addition, increasing the ratio of the Y element shifted the chromia film toward the amorphous state [16]. According to Salari Mehr et al. [27,28], the crystallinity of the atomic layer-deposited (ALD) chromia films greatly improved and the eskolaite phase formed with the increase of the substrate temperature from 200 to 275°C but deteriorated upon the addition of Ti. According to Joshi et al. [29], doping with La content \geq 4 at% La leads to the formation of a TiO₂ rutile structure as an additional phase alongside the anatase phase. In addition, loading of 0.1-0.3 at% La₂O₃ to ZnO-Bi₂O₃ microparticles through the flash-sintering route did not change the crystal structure (phase) of the varistor, and 0.2 at% La₂O₃ was enough to achieve a uniform structure with the highest density and best electrical efficiency [30]. These results indicate the possibility of controlling the chromia film crystallinity. The decrease in crystallinity correlated with an increase in defects and disorder within the materials. This will improve the film conductivity and optical characteristics, which will be covered in the following sections.

The analysis of the obtained data (the positions and relative intensities of the peaks) revealed that the formed material is a hexagonal chromia (α -Cr₂O₃, eskolaite phase) with a rhombohedral primitive cell (JCPDS nos 01-072-4555 and 01-084-0313 [26]). The Miller indices were inserted on all peaks. No trace related to La or its oxides was detected in the patterns. This outcome confirmed the purity of the product and aligned with the findings reported for La-

doped CuO [31] and Co₃O₄ [32]. La doping resulted in a peak shift that was observed for (0 1 2), (02 4), and (1 1 6) planes, indicating the successful replacement of Cr with La atoms. The smaller Cr atoms ($r_{\rm Cr} = 0.615$ Å) [33,34]) in the host lattice were substituted by larger La atoms ($r_{\rm La} = 1.19$ Å [34]), so that the host atoms suffered compression, and the lattice parameters contracted, resulting in a compressive strain [20].

The crystallite size C_s was determined using the wellknown Scherrer equation: $C_{s}(nm) = \frac{K \times 0.154}{FWHMx \cos \theta}$, where Scherrer's constant K = 0.89 and 0.154 for the wavelength of the used Cuk_a . The obtained C_s values are listed in Table 1. HT from 400 to 600°C resulted in an increase in the $C_{\rm s}$ value from 24 to 38 nm, while La doping reduced it to 16 nm at a content of 5.0% La. Similarly, loading of 15 and 30% Ba reduced the C_s of sol-gel-derived chromia from 24 nm to 15.7 and 15.2 nm, respectively [22]. The C_s determined for the spin-coated Cu_{0.94}Ni_{0.06}O and Cu_{0.88}Ni_{0.06}La_{0.06}O films were 24.2 and 11.35 nm, respectively [31]. Additionally, the $C_{\rm s}$ of the sol-gel-prepared (La₂O₃)_{99.5} (Cr₂O₃)_{0.5} composite increased from 10 to 30 nm with the increase of the HT temperature from 100 to 850°C [35]. In contrast to these results, the XRD patterns of the tapioca-assisted sol-gel-prepared chromia doped with Fe up to 18% showed no impurity peaks, and the $C_{\rm s}$ value increased from approximately 30 to 76 nm [33]. The C_s values of co-precipitated chromia decreased from 37.5 nm for the undoped sample to 28.1 nm at 2.0% Sr level but increased to 56.6 nm at 6.0% Sr content. Besides the Cr₂O₃ phase, the obtained chromia films were found to contain Cr₂O₅ and Cr₈O₁₃ phases. This indicates that the sol-gel/spin coating methods can provide pure films. Moreover, the Y/Cr₂O₃ multilayered films prepared by ionplating technology exhibited a (104) preferential growth. Some Cr and Cr₂O_{2.4} peaks were observed due to O₂ deficiency during film deposition, and the element Y was also detected when its content reached 12.23% [5].

The assessment of chemical (reactive) groups in the prepared chromia and La-doped chromia films was done *via* recording FTIR spectra in the wavenumber range $(400-4,000 \text{ cm}^{-1})$, as shown in Figure 2. The vibrations of metal–oxygen (Cr–O) bonds are prominent in the 400–1,000 wavenumber region because of the interatomic vibrations [3]. The inset of Figure 2 displays magnified vibration bands of chromia films annealed at 400°C (Cr/400) occurring at 500, 750, and 890 cm⁻¹. These peaks are assigned to stretching vibrations of Cr–O and Cr=O, as indicated in the figure. Besides, the less intense peak at 1,100 cm⁻¹ can be assigned to Cr–O– Cr. The HT of the films at 500 and 600°C resulted in wider bands with higher intensity, with a significant shift of the 500 and 750 cm⁻¹ bands to 486 and 765 cm⁻¹, respectively. However, the bands' intensity decreased upon La doping.

Investigating the exact position and intensity of Cr–O, Cr–O–Cr, and Cr=O vibrations is a controversial topic. In



Figure 2: Transmittance (FTIR) spectra of chromia films heat-treated at 400–600°C and doped with 2.5 and 5.0 at% La.

Table 1: Crystallite size (C_s), transmittance (T%), optical band gap (E_g), $\frac{N}{m^*}$ ratio, and lattice dielectric parameter (ε_L) of chromia films heat-treated at 400–600°C and doped with 2.5 and 5.0 at% La

Film	C _s (nm)	T (%)		E _g (eV)	$\left(\underline{e^2}\right)\left(\underline{N}\right)$	$\epsilon_{ m L}$	$R_{\rm sheet}$ (k Ω)
		at λ = 380 nm	at λ = 500 nm		$\left(\pi c^{2}\right)\left(m^{*}\right)$		
Cr/400	24	46	68	2.70	1.052 × 10 ⁻⁶	4.44	27.12
Cr/500	29	57	72	2.80	1.034 × 10 ⁻⁶	3.87	31.64
Cr/600	38	64	78	2.90	0.313 × 10 ⁻⁶	3.50	113
2.5% La-Cr/600	22	59	83	2.85	0.281 × 10 ⁻⁶	3.20	90.4
5.0% La-Cr/600	16	50	80.5	2.75	0.391 × 10 ⁻⁶	3.35	45.2

the Fe-doped Cr₂O₃ NPs prepared by the sol-gel route, Cr-O vibrational modes are located at 540 and 583 cm^{-1} [24]. Gyawali et al. [3] reported peaks for Cr–O bond stretching in the liquid-impregnated chromia NPs at 417 and 542 cm^{-1} . Alshammari and his group [22] noted peaks for Cr-O stretching in the sol-gel-prepared chromia NPs at 513 and 615 cm⁻¹. Also, they observed absorption peaks at 1,020 and 942 cm⁻¹, which arose from Cr–O–Cr and Cr–O vibrations, respectively. Ishtiag et al. [20] claimed that peaks for Cr-O stretching vibrations in the spray-deposited chromia and Cu-doped chromia films were observed at 611 and 827 cm^{-1} , while those for Cr=O and Cr-O-Cr were observed at 892 and 1,040 cm⁻¹, respectively. However, they detected vibrations related to C=O and H_2O molecules. In addition, Ashika et al. [10] detected small bands in the range of 1,450–4,000 cm⁻¹ attributed to the water and carbonate species that are adsorbed at the surfaces of chromia NPs. This again confirms that the sol-gel/spin coating methods yield chromia films with high purity. According to Khatamian et al. [26], the chromia NPs prepared by alkaline melting of the chromite ore exhibit two intense bands at 571 and 626 cm⁻¹ and less intense ones at 443 cm⁻¹, originating from the Cr–O stretching vibration. The peaks for Cr-O and Cr=O vibrations in the bio-synthesized chromia NPs were found at 610 and 836–901 cm⁻¹, respectively. Increasing/decreasing the sharpness and height of the peaks after HT/La doping is due to the improvement/ deterioration of the films' crystallinity [3]. The absence of La bands is due to the small concentration of La in the doped films. No new bands such as La-O or La-Cr were observed, which means that there is no chemical interaction between La and Cr₂O₃ films. These results align with the XRD findings and validate the ability to adjust the structure of chromia films through incorporation of HT and La.

The film surface morphology and the chemical (elemental) composition of the spin-coated chromia films on glass substrates annealed at 400°C (Cr/400) and 600°C (Cr/600) and 5.0% La-doped chromia (5.0 % La-Cr/600) were investigated by FE-SEM and the EDX, as illustrated in Figures 3(a)–(c) and 4(a)–(c), respectively. The Cr/400 film has grains of size in the range of 35–56 nm and a large number of small cracks.

These cracks may be formed during the pre-heating process due to solvent evaporation. Increasing the HT to 600°C resulted in grains with large sizes in the range of 45–72 nm and a reduced number of cracks. The 5.0% La doping had an impact on the film morphology, causing the grains to appear smaller and the zigzag (curved) cracks to have smaller widths. A similar morphology was observed for 2–10% La-doped TiO₂ [29]. This morphology increased the films' surface area for efficient performance in various



Figure 3: (a)–(c) FE-SEM surface images of Cr/400, Cr/600, and 5.0 %Ladoped Cr/600 films.

applications such as gas sensing, photocatalysis, and water treatment. A similar finding was reported for DC-sputtered chromia films, whose nanoflake-like morphology changed into flat-surface morphology after Pd inclusion [18]. The grain size (and morphology) evaluated by high-resolution transmission electron microscopy (HR-TEM) for the sol–gel-prepared Fe-doped Cr_2O_3 NPs was found to decrease from 36.56 nm (ellipsoidal shapes) to about 22.84 nm (nanorod-like) at 10 at% Fe, which then grew to 66.5 nm (spherical shape) at 30 at% Fe [24]. Huiming *et al.*

600 films.

[36] reported that La implantation at 3×10^{17} La ion/cm² on

the Co-40Cr alloy at 10^3 promoted the formation of the

and 5.0% La-Cr/600, respectively. These EDX spectra indi-

cate the existence of O and Cr peaks. Two peaks attributed

to Cr are observed at around 0.58 keV (Cr L_{α}) and 5.4 keV

(Cr K_{α}). The signal of O at 0.5 keV (O K_{α}) originates from the

Cr₂O₃ films and also from the glass slides. Additionally, the

Si (main component of glass), Na, Mg, and Ca signals are

from the glass used as substrates. The EDX has a volume of

interaction of about $\geq 1 \, \mu m$ (when the voltage of acceler-

ating is \geq 11 keV), which is larger than the film thickness

[37]. The spectrum of 5.0% La-Cr/600 contains three (L_{α} , L_{β} ,

and L_{ν}) lines that confirms the presence of La at 4.6, 5.4, and 5.8 keV, respectively. The ratio [Cr]/[O] in Cr/400 and

Cr/600 films appears to be the same. While the [Cr] is

decreased relative to [O] upon doping with 5.0 at% La.

Figure 4(a)–(c) shows EDX analyses of Cr/400, Cr/600,

Cr₂O₃ film with fine-grains.

of the Cr/400 film at these two wavelengths are 46 and 68%, which increase to 64 and 78% with the increase of the AT to 600°C, and then became 50 and 80.5%, respectively, after doping the Cr/600 film with 5.0% La. This result illustrates that the T% of the chromia films is adjustable, where it increases with annealing, due to the improved films' crystallinity, and is lowered by La incorporation, where the defects induced by La absorb or scatter the incident photons. Optoelectronic and photovoltaic devices can har-

The films' absorption coefficient (α) can be derived from T% using the following equation: $\alpha = \frac{1}{d} \ln (1/T)$, where *d* is the film thickness. In addition, the value of α is dependent on the photon energy (hv) through Tauc's relationship: $(hv \times \alpha)^{\frac{1}{r}} = A(hv - E_g)$, where A is a constant and $r = \frac{1}{2}$ for the direct optical transition. Figure 6 shows the plots of $(ahv)^2 v shv$, where extending the linear parts of the obtained curves to $(ahv)^2$ = zero (zero absorption) gives the values of E_g . The E_g of Cr/400 is found to be 2.7 eV, significantly increasing to 2.9 eV after annealing the films at 600°C. HT at high temperatures supply the atoms and molecules with the required thermal energy to occupy their normal (ideal) positions [38]. This increases the crystallinity and reduces the defects inside the material band gap. The E_g of the sputtered Cr_2O_3/CuO thin film at 100 W (200 W) increases from 2.33 (1.96) eV to 2.64 (2.47) eV under HT at increasing temperatures from 300 to 1,000°C. At 300 W, the $E_{\rm g}$ increases from 1.92 to 2.13 eV under HT at

Figure 4: (a)-(c) Elemental analysis of Cr/400, Cr/600, and 5.0 %La-Cr/

dopina.

90

80

70

3.2 Film transmittance and band gaps

The UV-vis-NIR transmittance (T%) spectra are shown in Figure 5. T% increases sharply in the UV region, and at $\lambda \geq$ 400 nm all films exhibit 60–90% transmittance. The inset of this figure is a magnified view for T% values in the 320-660 nm wavelength range. The values at 380 and 500 nm for all films are displayed in Table 1. The T% values







80



Figure 6: Tauc plot for E_{g} determination for the films under HT and La doping.

temperatures from 300 to 600°C and then decreases to 1.45 eV at 1,000°C [38]. At 300 W and 10,000°C, the high thermal energy may force the atoms to diffuse inside the substrate, which deteriorates the crystallinity.

La incorporation induces a reverse effect and narrows the E_g value from 2.9 to 2.75 eV. Doping with the La content produces defects or traps for the charge carriers and delays or inhibits their recombination, reducing the E_{g} value. Similarly, the $E_{\rm g}$ of ZnO decreases from 3.260 to 3.147 eV at 7.5 at% La content [25]. In addition, 0-12 at% Nb doping reduces the E_g of SnO₂ quantum dots from 4.76 to 2.18 eV [39]. In contrast, the $E_{\rm g}$ of the Cu_{0.94}Ni_{0.06}O film increases from 1.5 to 2.3 eV when the film is codoped with 6.0 at% La (Cu_{0.88}Ni_{0.06}La_{0.06}O) [31]. The transparency and $E_{\rm g}$ of materials are affected by materials' structural geometry [16]. The decrease in T% with increasing La content from 2.5 to 5.0 at% could be due to the introduction of more La in the lattice sites into interstitial positions [40]. Increasing the La content results in more defects, disorders, and O vacancies that absorb a portion of the incident light and reduce the transmitted part. In addition, the difference in ionic radii between the host and dopant atoms increases the defect sites and generates charge carriers inside the band gap.

The literature reported similar findings, showing that doping with 3.0% Ag reduces the E_g value of ion-impregnated chromia NPs from 3.4 to 2.82 eV, resulting from the defect sites formed in the composite films [3]. Values of 1.82

 $\leq E_g \leq 2.3 \text{ eV}$ were reported for chromia NPs prepared by hydrothermal, co-precipitation, thermal treatment, and ball milling under different synthetic conditions [10]. Fan et al. [12] reported that the PLD chromia thin films showed T% = 53–90% in the UV–vis spectra, and their E_g increased from 2.9 to 3.1 eV after doping with 5.0 at Ni. Bhardwaj et al. [16] reported a slight reduction in the $E_{\rm g}$ of chromia (from 3.61 to 3.56 eV) upon 5.0 at% doping with Ni, Mn, or Co, due to some structural distortions originating from the charge disproportion. Alshammari et al. [22] reported a band gap of 3.28 eV for the sol-gel chromia NPs, measuring 25 nm in size, which decreased to 2.8 and 1.5 eV upon loading 15 and 30% Ba. Chromia NPs prepared by the co-precipitation method showed a direct E_g of 3.3 eV, shifted toward 3.0 eV upon doping with 2, 4, and 6.0% Sr [23]. The band gaps of chromia and chromia/bentonite were found to be the same (3.09 and 3.08 eV) [26]. Addition of 45% Ti decreased the $E_{\rm g}$ of ALD chromia from 3.1 to 2.2 eV [27]. The direct (indirect) E_g of TiO₂ decreased from 3.38 (3.24) eV to 3.26 (2.53) eV at 4.0% La [29], then increased to 3.52 (3.02) eV at 10 % La. Tuning of this E_{σ} broadened the films' utilization for photovoltaic and optoelectronic devices [31]. The spray-deposited chromia films demonstrated 65% transmittance, which improved to 94% by In doping at 4.0 at% content. The $C_{\rm s}$ and $E_{\rm g}$ values of 2.0% In content increased from 23 nm and 3.6 eV to 29 nm and 3.93 eV upon increasing [In] to 4.0% and then decreased to 3.46 eV at 8.0% In [40]. Moreover, the E_g value of the spraypyrolyzed chromia decreased from 2.92 to 2.63 eV when doped with 4.0% Cu [41]. In contradiction to these results, Nikolaev et al. [4] reported an E_g of 3.06 eV for the undoped chromia layer grown by mist-CVD epitaxy, which then increased from 3.16 to 3.73 eV with increasing Ga content in the $(Cr_{1-x}Ga_x)_2O_3$ structure from x = 0.03 to 0.5. Loading 0–10 at% Cu through the microwave-assisted technique decreased $C_{\rm s}$ from 13.96 to 11.13 nm and increased the $E_{\rm g}$ of Cr_2O_3 NPs from 3.164 to 3.585 eV [42].

3.3 Chromia films' optical constants

Evaluating optical constants such as the absorption index $(k, \text{ where } k = \frac{\lambda \times a}{4\pi})$ and refractive index (n) are essential for optical communication and optoelectronic devices. Figure 7 shows the *k* distribution in the λ range of 280–1,750 nm. First, *k* values are high in both the UV and NIR regions, compared with those in the visible wavelengths. Second, Cr/400 exhibits two bands at 465 and 625 nm, which are related to the six-coordinated ${}^{4}A_{2g} \rightarrow {}^{4}T_{1g}$ and octahedral ${}^{4}A_{2g} \rightarrow {}^{4}T_{2g}$ of d^{3} intrinsic electronic transitions for Cr³⁺ [43]. The *d*-*d* transitions from t_{2g} (triple





Figure 7: Absorption index dependence on the HT (400–600°C) and La content in chromia films.

degenerate) to $e_{\rm g}$ (double degenerate) levels originate for these peaks and induce the green color of chromia [10]. Third, the absorption edge at 370 nm signifies the E_{g} of chromia [23]. This edge becomes clearer with the increase of the temperature from 400 to 600°C and becomes a band with enhanced intensity upon doping with La. In other words, the edge (band) at 370 nm grows in intensity at the expense of the other two bands at 465 and 625 nm. Fourth, the k values decrease with increasing AT and then increase with increasing La content from 0 to 5.0 at%. The behavior of *k* is related to the change in the film surface morphology, the increase/decrease in the crystallinity, and the available number of charges to absorb the incident light. This finding suggests that chromia-based IR sensors could be developed by controlling the AT and La content.

Figure 8 shows the dependence of the films' reflectance (*R*%) on the temperature and La doping. Materials with a larger number of particles exhibit higher reflectivity. Increasing the AT from 400 to 600°C results in increasing particle (grain) size, see Figure 3(a) and (b), where the particle tends to coalesce and thus reduces the number of particles. However, La doping leads to a reduction in the particle size, as revealed by XRD and SEM images. La-doped chromia appears with a large number of particles, each particle acting as a scattering or reflecting center. The *n* values are derived from *R* and *k* using the relation: $n = \frac{\sqrt{4R - (1 - R)^2k^2 + R + 1}}{(1 - R)}$ [44]. The inset of Figure 8

Figure 8: Dependence of reflectance R (%) and refractive index n (the inset) on HT and La content.

shows the *n* vs λ plot in the visible and NIR (400–1,350 nm) spectral regions. The *n* values of Cr/400 of 1.97–2.27 decrease to 1.85–1.9 with increasing AT to 600°C. The *n* values of the 2.5% La-Cr/600 film improve from 1.77–1.85 to 1.80–1.925 as the La content is increased. For all films, the highest *n* values correspond to λ equal to 525 and 670 nm. This behavior may be due to the fact that the incident photon frequency is equal to the plasma frequency [18]. The higher values of *n* for Cr/400 are due to the limited crystallinity and existing traps and disorders, which cause a higher level of photon scattering and reduced T%. These results are consistent with XRD and FE-SEM analyses.

Another two optical constants, the lattice dielectric constant ($\varepsilon_{\rm L}$) and ($\frac{\text{carrier concentration}N}{\text{electron effective massm}^*}$), can be determined from the dispersion relation: $n^2 = \varepsilon_{\rm L} - \left(\frac{e^2}{\pi c^2}\right) \left(\frac{N}{m^*}\right) \lambda^2$ [42], where *e* and *c* are the electron charge and light speed, respectively. Figure 9 shows the $n^2 vs \lambda^2$ plots for all films. Extending the curves to meet the *y*-axis gives $\varepsilon_{\rm L}$ values, and the slope represents $\frac{N}{m^*}$ in terms of $\frac{e^2}{\pi c^2}$. The deduced values of $\frac{N}{m^*} \left(\frac{e^2}{\pi c^2}\right)$ and $\varepsilon_{\rm L}$ are given in Table 1. With the increase of the AT, both $\varepsilon_{\rm L}$ and $\frac{N}{m^*}$ decrease from 4.44 and 1.052 × 10⁻⁶ to 3.5 and 0.313 × 10⁻⁶, respectively. When the Cr/600 film is doped with 5.0% La, the $\frac{N}{m^*}$ with AT/La doping is consistent

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Figure 9: $n^2 vs \lambda^2$ plot for determining the $\frac{N}{m^*}$ and ε_L optical parameters.

Figure 10: I–V characteristic curves of chromia films annealed at different temperatures and doped with La.

with the increase/decrease of the E_g of the films. The obtained results confirm that the optoelectronic properties of the chromia films can be controlled by changing the experimental conditions (HT) and doping approach. The chromia films are therefore the best candidates for designing optical filters with a reasonable E_g .

3.4 I–V characteristic curves

Figure 10 shows the I–V plots of the spin-deposited chromia films on the glass substrate. The linear I–V curves indicate good Ohmic resistance. Bhardwaj et al. [16] also found a linear I–V relationship for chromia doped with Ni, Co, and Mn that was deposited on Al₂O₃ substrates. The films' sheet resistance (R_{sheet}) was calculated from the slope ($\frac{\Delta I}{\Delta V}$) of the equation: $R_{\rm sh} = \frac{\pi}{\ln 2} \frac{1}{\text{slope}} = \frac{4.532}{\text{slope}}$ [41], where 4.532 is the correction factor. The R_{sheet} values, Table 1, are high $(27.12-113 \text{ k}\Omega)$. These values are higher than those reported for the spray pyrolyzed films [40]. The smallest R_{sheet} value was exhibited by the film that was annealed at 400°C. The R_{sheet} value increased with increasing AT, which then decreased upon La loading. During annealing, the atoms and molecules get sufficient thermal energy for more arrangement and occupying the ideal positions. However, La introduces more disordering and defect states in the material's band gap. Therefore, this change in R_{sheet} is consistent with the variation in films' $E_{\rm g}$ values. The resistivities of the spray-deposited chromia and 4.0% Cu-doped chromia films were found to be 960 and 33.06 Ω cm, respectively [41]. This variation in R_{sheet} values is related to the change in films' crystallinity, morphology, and chemical composition.

4 Conclusions

The influence of HT at temperatures in the range of 400–600°C and La incorporation on the physical properties of chromia thin films spin-deposited on glass substrates was investigated. XRD showed that hexagonal chromia was formed after annealing at 400°C. Raising the temperature to 600°C enhanced the films' crystallinity and increased the $C_{\rm s}$ value from 24 to 38 nm, which then dropped to 16 nm upon doping with 5.0% La. The Cr–O and Cr=O stretching vibrations of chromia were detected at 500, 750, and 890 cm⁻¹. Increasing the AT (increasing the La content) resulted in a small shift of these bands, and their intensity increased (decreased) based on the changes in the films' crystallinity. FE-SEM revealed that the films are formed from grains separated by a large number of small cracks. EDS spectra confirmed the existence of La in the film composition. UV-vis-NIR spectra revealed that HT in the range of 400-600°C improved T% from 46 to 64%, which then decreased to 50% due to La doping. The $E_{\rm g}$ increased from 2.7 to 2.9 eV and then reduced to 2.75 at 5.0% La. The films showed high absorption indices in the

UV and NIR regions. The R% and n oscillated between 6 and 15% and 1.77 and 1.97, respectively. In addition, the $\varepsilon_{\rm L}$ of the films was in the range of 3.2–4.44, and $\frac{N}{m^*}(\frac{e^2}{\pi c^2})$ was found to decrease from 1.052 × 10⁻⁶ to 0.313 × 10⁻⁶ after HT and increase to 0.391×10^{-6} upon addition of 5.0 at% La. The films displayed linear I–V curves (good Ohmic resistance) and R_{sheet} in the range of 27.12–113 k Ω . In summary, it is possible to engineer the films' crystallinity, morphology, band gap, optical constant, and resistance of the chromia films by HT and La content. Tuning the structure and optical and electrical properties of the chromia films widen their utilization in the practical and technological applications in the fields of optoelectronics, photonic devices, and chromia-based IR sensors. However, more studies are required on chromia films on Si or sapphire substrates for HT at higher temperatures than 600°C and the possible device fabrication and characterization.

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