





Course: English Language









A. Basic Information

Program(s) on which the course is given:

Department offering the course

Pharmacognosy

Faculty offering the program

Arts

Clinical

Dept. responsible for teaching the course

English department

Academic year / level

lst

Course title

English Language

Course code

EN 101

Contact hours (credit hours)

2...

Pre-requisite of the course:

No

Course coordinator

Dr. Hala Rashed

Major or Minor element of program

Minor

Date of specification approval

09/09/2015

B. Professional Information

1. Overall Aims of Course

Training in reading, comprehension, basic grammatical rules, writing and translation. The course adopts a systematic approach to proper essay writing, such as idea development, paragraph structure, introductions, support, and conclusions.

- 2. Intended Learning Outcomes of Course (ILOs)
- a- Knowledge and Understanding:

By the end of the course, the students should be able to:

- a1. Read and write scientific and non-scientific articles
- a2. Apply basic grammatical rules

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b- Intellectual Skills

b1. Adopts a systematic approach to proper essay writing

b2. Integrate basic grammatical rules

By the end of this course, the student should be able to:

c- Professional and Practical Skills

c1. Utilize proper medical and pharmaceutical terms By the end of the course, the student should be able to

c- General and Transferable Skills

By the end of the course, the student should be able to:

d1. Communicate verbally using idea development, paragraph structure, introductions, support, and conclusions.

Contents	The state of the s		
Teaching	TOPIC	No, of lecture	No. of Practical
week		hours	hours
1	Training in reading		
2	Training in reading	2	0
3	Training in comprehension	2	0
4	Training in comprehension	2	0
5	Training in basic grammar rules	2	99
6	Training in basic grammar rules	. 2	0
7	First periodic exam	2	0
8	Training in writing and translation	2	0
9	Training in writing and translation	2	0
10	A systematic approach to proper essay writing, such as idea	2	0
	A systematic approach to proper essay	2	0
11	writing, such paragraph structure, introductions, support, and conclusions.		<u> </u>
12	Second periodic exam	2	0
Total no of	24		







		 	 AULU AULU!		 · ·		
٠	hours					*	
	13		 FINAL Exam	-	 the second		
	Marine Commence		E STANL LACINE				

4. Teaching and Learning Methods

- 4.1- Lectures (board, data show)
- 4.2- Assignments
- 4.3- Class discussion

5. Student Assessment Methods

- 5.1. Practical exams to assess knowledge and understanding as well as intellectual skills.
- 5.2. Written exams to assess all types of skills and mainly general and transferrable skills practice.

Assessment Schedale

Quiz 1

Quiz 2

Final exam

Weighting of Assessment Periodical

Final exam

Total

6. List of References

- 6.1- Course Notes: Lecture notes in English a grage
- 6.2- Essential Books (Featbooks)
 - 1. English for Academic Research: Writing Exercises by Adrian Wallwork

Facilities required for teaching and learning

- 1. Lecture rooms with data show
- 2. Procurement of latest edition of the above-mer time texts and there to update the education process

Course Coordinator: Dr. Hala Rashed

Head of Department: Prof. Mona Hetta

Date: 09 /09/2015

cording to semester schedul







Course: Mathematics and Statistics

Course code: MS 101









A. Basic Information

Program(s) on which the course is given:

Clinical

Department offering the course

Organic and pharmaceutical chemistry

Faculty offering the program

Pharmacy

Dept. responsible for teaching the course Mathematical science

Academic year / level. First level / first semeste

Course title: Mathematics and Statistics

Course code: MS 101

Contact hours (credit hours): 1

Pre-requisite of the course: Non

Course coordinator: Dr. Maged Kassab

Major or Minor element of program. Minor

Date of specification approval 09/09/2015

B. Professional Information

The course aim and intended learning outcomes are based on that mentioned in the programme specifications, with more course-related specific details.

1. Overall Aims of Course

The aim of the course is to provide students with a detailed knowledge and understanding of functions and graphs, limits and continuity differentiation, exponential, logarithmic and trigonometric functions, integration, basic differential equations, function of several variables and problems related to them. The course also provides the students necessary knowledge of the probability, random variables and correlation and regression as well as hypothesis testing.

2. Intended Learning Outcomes of Course (ILOs)

a- Knowledge and Understanding:

By the end of the course, the students should be able to:

Theme of theme







- a1. Understand the theoretical mathematical principal of functions and graphs
- a2. Understand of the differentiation, exponential, logarithmic and trigonometric functions.
- a3. Identify the concept of integration, basic differential equations.
- a4. Recognize the concept of hypothesis testing.

b- Intellectual Skills

- b1. Perform different mathematical and statistical calculations
- b2. Distinguish the different type of functions and equation differentiation and integration
- b3. Solve some problems encountered in hypothesis testing By the end of this course, the student should be able to:

c- Professional and Practical Skills

- c1. Formulate different mathematical and statistical calculations
- c2. Evaluate a successful formula related to different equation differentiation and integration
- c3. Design some probability and hypothesis testing By the end of the course, the student should be able to:

d- General and Transferable Skills

By the end of the course, the student should be able to:

- d1. Work independently in different mathematical and statistical calculations software
- d2. Apply problem-solving skills for the mathematical and statistical calculations within the practice of pharmacy fields.
- d3- Develop effective communication skills and critical thinking

The aim of the course is to provide students will a detailed knowledge and understanding of functions and graphs, limits and continuity, differentiation, exponential, logarithmic and trigonometric functions, integration, basic differential equations, function of several variables and problems related to them. The course also provides the students necessary knowledge of the probability and random variables as well as hypothesis testing.







3. Contents

Teaching week	TOPIC	No. of lecture hours
1	Functions and Graphs 1	1
2	Functions and Graphs 2	1
3	limits and continuity	1
4	Differentiation 1	1
5	Differentiation 1	1
6	Quiz 1	1
7	Integration 1	1
8	Integration 2	i
9	Quiz 2	1,
10	Probability	- 1
11	Correlation and regression	1
12	Hypothesis testing 1	1
13	Hypothesis testing 2	1 7
Total no of hours		and the state of t
14	FINAL Exam	

4. Teaching and Learning Methods

- 4.1- Lectures (board, data show)
- 4.2- Assignments
- 4.3- Class discussion

5. Student Assessment Methods

- 5.1. Written exams to assess knowledge and understanding as well as intellectual skills.
- 5.2. Worksheets and assignments to assess all types of skills and mainly general and transferrable skills practice.

Assessment Schedule

Quiz 1

Quiz 2

Practical exam

Final exam

4th or 5th week

14th week; according to semester schedule







3. Contents

Teaching week	TOPIC	No. of lecture hours
1	Functions and Graphs 1	1
2	Functions and Graphs 2	1
3	limits and continuity	1
4	Differentiation 1	1
5	Differentiation 1	1
6	Quiz 1	
7	Integration 1	1
8	Integration 2	Ĭ
9	Quiz 2	1
10	Probability	1
11	Correlation and regression	1
12	Hypothesis testing l	1
13	Hypothesis testing 2	1
Total no of hours		
14	FINAL Exam	

4. Teaching and Learning Methods

- 4.1- Lectures (board, data show)
- 4.2- Assignments
- 4.3- Class discussion

5. Student Assessment Methods

- 5.1. Written exams to assess knowledge and understanding as well as intellectual skills.
- assess all types of skills and mainly general and 5.2. Worksheets and assignments to transferrable skills practice.

Assessment Schedule

Quiz 1

Quiz 2

Practical exam

Final exam

14th week; according to semester schedule







Weighting of Assessments

Periodical 10%

Worksheets and assignments 15%

Final exam 75%

Oral exam

6. List of References

Rouaud, Mathieu. (April 2017) [First published July 2013]. Probability, Statistics and Estimation (PDF). p. 10. Archived (PDF) from the original on October 9, 2022. Retrieved February 13, 2024.

- 6.1- Course Notes: Lecture Notes
- 6.2- Essential Books (Textbooks)
- 6.3- Periodicals:
 - # 6.4- Web Sites:

Facilities required for teaching and learning

- 1. Lecture rooms with data show
- 2. Procurement of the latest edition of the above-mentioned texts and others to update the education process

Course Coordinator: Dr. Maged Kassab

Head of Department: Prof. Mana Hetta

Date: 09 /09/2015







Course title: Cell Biology

Course code: MD 102

A. Basic Information:

Course Title:	Cell Biology				
Course Code:	MD 102				
Program on which the	Clinical				
Department offering th	Biochemistry				
Academic year/ level:	cademic year/level: 1st Semes		Level:1		
Prerequisite:	Organic chemistry			****	
Credit hours:	Lecture:1	Practical:	1	Total:2	

B. Professional Information

1. Course Aims:

The course provides the basic principles of cell biology, classification of prokaryotes and eukaryotes with an emphasis on cellular structure. It describes the functions of organelles and regulatory mechanisms in animal cells

2. Intended Learning Outcomes (ILOs):

a. Knowledge and understanding:

At the end of this course, student should be able to:

Al al-	Identify the fundamental concepts of biology, with an emphasis on animal cell
	structure and function.
	Describe the basic characteristics of prokaryotic and eukaryotic cells and
	extracellular matrix
A4 a3.	Enumerate the theories of transport across the biological membranes
1 a4,	Recall the basic principles of cellular function including cell communication.
(FREE CASTES	Recall the basic principles of cellular function including cell communication, ligands and receptors, DNA and RNA structure and function
a5	Describe gene expression process mechanism and regulation.

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b. Intellectual Skills:

At the end of this course, student should be able to:

	Correlate cell structure with its function, including active and passive transport
bl.	across membrane. Also, nucleic acid role in protein synthesis.
B18 b2.	Integrate basic biochemical facts with cell cycle, cell division and mechanisms of
	apoptosis in normal cells and tumor formation.
	Investigate the characteristics of living things to classify organisms into eukaryotes
	and prokaryotes.

c. Professional and Practical Skills:

At the end of this course, student should be able to:

TABLE REPORT OF THE PROPERTY O	Utilize gained information to communicate with other health care professionals
C11 c2	Apply biology knowledge to practical scientific activities and relevant applications
	of biology in society and the environment.
· © 14+ •3.	Conduct research studies including presentation skills and discussion

d. General and Transferable Skills:

At the end of this course, student should be able to:

D3 d1,	Work effectively as a part of a team to perform the required tasks
D4 d2.	Develop the skills required for continued self-professional development and self-learning

3. Contents:

3.1. Lectures:

Study week	Topies	No. of Credit Hours
1.	Prokaryotic and Eukaryotic cell structure and functions	1
2.	Animal cell; structure and functions	1
3.	Biological membranes structure and function	1
4.	ECM and cell movement	1
5.	Cytoskeleton	1
6.	Cell cycle and cell division	1
7.	cellular bioenerge	1
8.	cell signaling and communication	1
9.	DNA structure and function	1
10.	RNA structure and function	1







11	protein synthesis	1
12	Control of gene expression	1
13. 14.	Final Exam	Total; 1 credit hours

3.2. Practical:

Study week	Topics	No. of Credit Hours
1.	Lab safety and microscope	
2.	Cell structure	1
3.	Cell membrane	1
4.	Nucleus	1
5.	Golgi apparatus	1
6.	Endoplasmic reticulum	1
7.	Endoplasmic reticulum	1
8.	Vesicles	1
9.	Lysosomes	1
10.	Cell cycle	1
11.	Cell division	
12	Revision	
13.	Final Exam	Total: 1 credit hour

4. Teaching and Learning Methods:

4.1.	Lectures	
4.2.	Practical lab	The same
4. 3	Research in library and web (homey	vonk & assignments)
4.4	E-learning ·	

5. Student Assessment Methods:

5.1. Assessment methods:







1. Written exam	to assess knowledge, understanding, intellectual and professional skills.	
2. Practical exam	to assess professional and practical skills.	
3. Course work	to assess knowledge, understanding, intellectual skills, general and transferable skills.	
4. Oral exam	to assess knowledge, understanding, intellectual skills, general skills and confidence.	
5. Quizzes	to assess knowledge, understanding and intellectual skills.	

5.2 Assessment schedule:

Assessment 1	Quiz 1	4 th week
Assessment 2	Mid-Term	8 th week
Assessment 5	Practical exam	11 th week
Assessment 6	Oral exam	12 th & 13 th weeks
Assessment 7	Written exam	12 th & 13 th weeks

5.3 Weighing of Assessments:

1. Course work:	
- Quiz 1	5
- Mid-Tern	. 10
2. Final-Term Exam	. 50
3. Oral Exam	10
4. Practical Exam	25
Total	100%

6. List of References:

No.	Reference	Туре
1.	Lippincott Illustrated Reviews: Cell and Molecular Bollow, by Nalini Chandar and Susan Viselli 1st Edition (Lippincott Mustrated Reviews Series, 2011)	Textbook
2.	Cell Biology and Histology by Gartner, Leslie P., Hiatt, James L., Strum, Judy, 5th Edition [Lippincott Williams & Wilkins, 2006]	Textbook





Journal of biology (h	ttps://jbiol.biomedcentral.com/
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- Current Biology (<u>https://www.cell.com/current-biology</u>)
- Cell reports (http://www.cell.com/cell-reports)

Periodical

7. Matrix of course contents versus ILOs:

7.1. Lectures:

dy ek	Course Contents	ILOs
	Prokaryotic and Eukaryotic cell structure and functions	a1, b1, b3, c1,c2
	Animal cell; structure and functions	a2,b3, c1
		a1, a3, b1 ,b3, c2, c3, d1
•	Biological membranes structure and function	d2
	ECM and cell movement	a3, b1, c1, c2, d2
	Cytoskeleton	a3, b1, c1, c2, d2
	Cell cycle and cell division	a5, b2, c2, c3, d1, d2
	cellular bioenergetics	a4, b1, c2, d2
 3.	cell signaling and communication	a4, b1, c2, d2
),	DNA structure and function	a4, b1, c2,c3, d2
0.	RNA structure and function	a4, b1, c2,c3, d2
1	protein synthesis	a4, b1, c2,c3, d2
12	Control of gene expression	a4, b1, c2,c3, d2

7.2. Practical:

tudy	Course Contents LLOs	
veek	Lab safety and microscope all all, b1, b3, c1,c2	JW 1629.
1.	Lab safety and interessors	







(2013 2010)		
2.	Cell structure	a2,b3, c1
	Cell membrane	a1, a3, b1 ,b3, c2, c3, d1
3.		d2
4.	Nucleus	a3, b1, c1, c2, d2
5.	Golgi apparatus	a3, b1, c1, c2, d2
6.	Endoplasmic reticulum	a5, b2, c2, c3, d1, d2
7.	Endoplasmic reticulum	a4, b1, c2, d2
	Vesicles	a4, b1, c2, d2
8.	Lysosomes	a4, b1, c2,c3, d2
9.		
10.	Cell cycle	a4, b1, c2,c3, d2
11.	Cell division	a4, b1, c2,c3, d2
	Revision	a4, b1, c2,c3, d2
\		

	Name Signature
Course Coordinator:	Dr. Nada Osama
Head of Department:	Prof. Mona Hetta
Approval Date	(数 / 109109/2015 ··
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Course: Biophysics

Course code: MD 101









A. Basic Information

Program(s) on which the course is given:

Clinical

Department offering the course

Pharmacology and Toxicology

Faculty offering the program

Sciences

Dept. responsible for teaching the course

Biophysics

Academic year/level

1st

Course title

Biophysics

Course code

MD 101

Contact hours (credit hours)

2

Pre-requisite of the course:

No

Course coordinator

Prof. Abdelmohsen pasha

Major or Minor element of program

Minor

Date of specification approval

09/+9/2015

B. Professional Information

1. Overall Aims of Course

Cell membrane structure, method of transport, channel types, receptors. Application of action potential, electrocardiogram, and electroencephalogram identification and wave's elucidation

- 2- Intended Learning Outcomes of Course (ILOs)
- a- Knowledge and Understanding:
- a1. Identify different cell membrane structures, method of transport, channel types and receptors.
- a2. Identify electrocardiogram and electroencephalogram
- a3. Discuss different applications of action potential

By the end of the course, the students should be able to

b- Intellectual Skills

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- b1. Integrate basic biological and physical sciences to understand different body functions.
- b2. Identify different techniques to evaluate different body functions

By the end of this course, the student should be able to:

c- Professional and Practical Skills

c1. Utilize different techniques to evaluate several body functions like electroencephalogram and electrocardiogram.

By the end of the course, the student should be able to:

c- General and Transferable Skills

By the end of the course, the student should be able to:

d1. Communicate verbally and problem solving

Contents

			1 1 11
Teaching week	TOPIC	No. of lecture hours	No. of Practical hours
1	Introduction to biophysics		1
2	Cell membrane structure		1
3	Cell membrane structure		1
4	Method of transport	1	
5	Channel types	100	1
6	Receptors		1
7	First periodic exam		1
8	Action potential	1	1
9	Electrocardiogram .	1	1
10	Electroencephalogram	1	1
11	Different wave's elucidation.	1	1
12	Second periodic exam	1	11
Total no of	24		<u> </u>







hours	(2013 – 2010)
13	
	FINAL Exam
Topobing	

2. Teaching and Learning Methods

- 4.1- Lectures (board, data show)
- 4.2- Assignments
- 4.3- Class discussion

3. Student Assessment Methods

- 5.1. Practical exams to assess knowledge and understanding as well as intellectual skills.
- 5.2. Written exams to assess all types of skills and mainly general and transferrable skills practice.

Assessment Schedule

'Quiz 1

7th week

Quiz 2

12th week

Final exam

13" week; according to semester schedule

Weighting of Assessments

Periodical

10%

Practical

25%

Final exam

65%

Total

100%

4. List of References

- 6.1- Course Notes: Lecture notes in biophysics
- 6.2- Essential Books (Textbooks)
- 1- Fundamentals of Biophysics by Andrey B. Rubin: Wiley & Sons, Incorporated, John

Facilities required for teaching and learning

1. Lecture rooms with data show

Course Coordinator: Prof. Abdelmohsen pasha

Head of Department: Prof. Mona Hetta

Date:

09/9/2015















Medicinal plants PC101









Course Specifications

A- Basic information

Program(s) on which the course is given

Bachelor of clinical pharmacy

Department offering the course

Pharmacognosy department

Faculty offering the program

Faculty of Pharmacy, Fayoum University

Dept. responsible for teaching the course

Pharmacognosy department

Academic year / level

First level, first semester

Course title

Medicinal plants

Course code

PG 101

Contact hours (credit hours)

Lecture 2 (2) + Practical 2 (1): Total 4 (3)

Pre-requisite of the course:

Registration

Course coordinator

Prof. Mona Hetta

Major or Minor element of program

Major

Date of specification approval

09/9/2015

B-Professional Information

1-Overall aims of the course:

Upon successful completion of this course, the students will gain the basic knowledge about the process of commercial production of medicinal plants including; cultivation, drying, packing and authentication. Students will also be able to demonstrate basic concepts of pharmacognosy and relate medicinal actions of some medicinal plants composed of leaves to their content of secondary metabolites

2-Intended learning outcomes (ILO's):

a-Knowledge and Understanding:

By the end of this course, the student should be able to: & al.definePharmacognosy and other botanical terms.

a2. Describe herbal drugs according to botanical origin, chemical onstituents medicinal activity, and difference between wild and cultivated plants

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- . Mention different methods of cultivation, collection, drying, packing and storage, ulteration of medicinal plants.
- . Memorize different classes of secondary metabolites and their chemical identification.
- . Identify natural drugs, especially leaves, macro-, micro-morphologically, emically, and animal drugs.b-Intellectual Skills:
- the end of this course, the student should be able to:
- . Apply different concepts about plant cultivation, drying, storage, and adulteration...
- . Compare between different secondary metabolites.
- . Recognize the difference between various morphological characters that able students to recognize medicinal leaves.c-Professional and Practical tills:
- the end of this course, the student should be able to:
- . Handle microscope and chemicals.
- . Examine different dusting powders microscopically.
- .Demonstrate unknown plant leaves microscopically

General Skills:

the end of this course, the student must be able to:

- .Work in a team to differentiate between several types of medicinal plants.
- .Deal with internet to collect data and prepare a research project about different natural proroducts.



Teaching week	ТОРІС	No. of lecture credit hours	No. of practical credit hours
1	Introduction of medicinal plants (definitions, etc),	2	
1	Lab safety		1
2	Cultivation of medicinal plants, environmental condition, soil composition, growth regulators	-2005-	
	Microscopy, how to make a mount (mount in water, iodine). Collection, drying of crude drugs, different types		1
3	Collection, drying of crude drugs, different types of adulteration and their detection	2	
3	Starches		P.
4	Packing and preservation of drugs, plant growth regulators.	2	
	Dusting powders Secondary plants metabolites (simple phenolics) flavonoids, Alkaloids)	2	1
.	Isolated elements, Introduction to leaves + Senna (Powder)		1
	Secondary plant metabolites (glycosides)	2	eseana.
0	Solanaceous Introduction, Belladonna (Powders)		1
7	Taxonomy	2	
	Datura + Hyoscyamus (Powder)		1
8	Plant cytology		
	Digitalis (Powder) Introductions of drugs composed of leaves (Leaf	2	
9	morphology, lamina) Revision (Starch & Dusting)	addition of the same of the sa	1
10	Sonna leaf	2	
	Revision (Vegetable powder) Solanaceous leaves:		1
11	Belladonna, Stramonium, Hyoscyamus	10 10 mm	
12	Digitalis, Henna and other medicinal dea	2 %	1
Total no of hours			11

Teaching and Learning Methods (lectures, open discussion, role plays, ..etc):

1 Interactive Lectures

(5102 - 5102)Course Specifications

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- 4.2 Presentations, discussions and seminar
- 4.3 Research in library and web.
- 4.4 Practical laboratory work

5- Student Assessment:

Assessment Schedule:

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2 ziuQ

Final written exam Final practical exam

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Assessment method

I ziuQ

2 zinQ

Activities

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Final written exam

Oral exam

6-List of References:

Course notes: 1.9

(textbooks): Essential books 2.9

pooks: Recommended €.3

Periodicals, Web

sites, Etc:

Facilities required for teaching and learning:

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www.herb.com

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www.herbdatanz.com/index.htm

www. Bestnetcraft.com

www.who.int/medicines/libraty/um/medicinalplants

Egyptian Pharmacopoeter Miliah, ext (2005).

Textbook of Pharmacognosy, CBS Publishers.

Toronto, 16th edition Elservier, 2010

Fundamentals of Pharmacognosy and Phytotherapy, 2 Heinrich, M., Barnes, J., Cibbons S. and E.M. Williamso

Narayana, P.S., Pullaiah, T., Varalakshmi, D., Vol 2 (201

London, New York, Oxford, Philadelphia, St. Louis a

Evans, W.C., Trease and Evans Pharmacognosy, Edinburg

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 $10_{\rm pp}$ week

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Lecture halls







- Laboratories, supplied with data show.
- White board.
- Books, library furnished with textbooks.
- Genuine plant drugs.
- Optical microscopes in labs.
- Slides and covers.
- Chemicals and reagents.
- Grinder.

Teaching Staff: Dr. HossamMokhtar – Prof, Mona Hetta

Course Coordinator: Prof. Dr. Mona H. Hetta

Head of Department: Prof. Dr. Mona H. Hetta

Date: 09/9/2015

Course specification

ourse code:	PC 103		
ourse name:	Pharmaceutical Organic		
	Chemistry I		
redit hours of the course:	Lecture: 2		
	Practical: 1		
	Total: 3		
re-requisite of the course:	None		
epartment teaching the	Pharmaceutical Medicinal		
ourse:	Chemistry Department		
rogram for which the course given:	Clinical Pharmacy Program		
ourse Co-ordinator:	Dr Mohammed Ibrahim		
	Abdel_Latif Hamed.		
•	Dr/ Mohammed Mahmoud		
	El_Agawany		
ead of the Department:	Prof. Dr./ Mona Hetta		
ate of specifications pproval	.9/٢.15		

professional information

)verall aims of the course:

on successful completion of this course, the students will develop various competencies based on vering the Sawing general outlines:

o make student think logically about atomic and molecular properties in addition to their interactions. o enable students to know how to deal with the reactions of organic compounds, the investigation of air properties and the relation between their molecular structure and their reaction mechanism. o prepare student with a sound understanding of the fundamental principles of organic and armaceutical compounds with respect to related subjects next years.

ntended learning outcomes (ILO's):

owledge and Understanding:

the end of this course, the student should be able to:

Know the basic concepts of aromaticity and chemistry of arenes

- -Know the characteristics of chemistry of benzenoids and polynuclear compounds
- -Understand The basics of spectroscopy and spectral analysis of organic compounds

tellectual Skills:

- the end of this course, the student should be able to:
- Recognize organic compounds from their physical and chemical properties.
- Discuss organic problems of different organic classes.
- Apply suitable methods for synthesis, purification, and identification of chemical substances from

fferent origins.

- Analyze and interpret experimental studies as well as data from published literature using ectroscopy.

Professional and Practical Skills:

the end of this course, the student should be able to:

- -Select the appropriate methods for identification and synthesis of organic compounds
- Identify aromatic organic compounds and carbohydrates physically and chemically.
- Analyze spectral data of organic compounds.

General Skills:

y the end of this course, the student should be able to:

- 1- Use the literature via different information sources including the library, internetetc.
- 2- Care about himself, his colleagues and all the tools and instruments of his lab during dealing with hemical organic compounds

3-course contents

	lecturer	lecture	practical	total
opics	Dr/Mohammed Ibrahim	2	1	1
ntroduction to Organic Chemistry-1	DI/Wollammed iolamm	_		
ab safety rules and Policies, systematic	,			<u> </u>
dentification of an organic compound	Dr/Mohammed Ibrahim	2	1	1
Alkanes, cycloalkanes, & alicyclic	DIAMOUAHILIEG in ariini	-	_	
compounds: Preparation & Reactions	Dr/Mohammed Ibrahim	2	1	1
Alkenes & Alkynes Preparation & Reactions	Dr/Monammed ibrahim			
Isomerism & Stereochemistry,	Dr/Mohammed Ibrahim	2	1	1
Stereoisomers			1	
ChemSketch Drawing Software			<u> </u>	
Molecular representation, Numbering &	Dr/Mohammed Ibrahim	2	1	1
Naming of Stereoisomers				<u> </u>
Compound with 2 or more chiral atoms,	Dr/Mohammed Ibrahim	2	1	1
chiral cyclic compounds, racemization				
and resolution				<u> </u>
	Dr/Mohammed El	2	1	1
Conformational Analysis	Agawany			1
Melting point determination	All the state of t			<u> </u>
Aromaticity & Aromatic ions	Dr/Mothammed El	<u>^</u> 2	1	1
Recrystallization	Agawapy	J 357		
Aromatic substitution reactions &	De Mahammed El	\2	1	1
	A CONTROL MANAGEMENT	A) -		
directing group		₩		
Recrystallization				
	6			

Introduction to mass spectro fragmentation and Mc-Lafferty rearangment Introduction infrared, Hook low, region of	Dr/Mohammed El Agawany2	2	1	1
different type of function groups. Introduction to NMR, type uses in life. H-	Dr/Mohammed El Agawany	2	1	1
NMR. Chemical Shift and TMS. Chemical shift of different type of protons. "C-NMR chemical shift. Problems on C-NMR)	Dr/Mohammed El Agawany	2	1	1
otal		24	12	36

4-Teaching and Learning Methods (lectures, open discussion, role plays, etc):

- 4.1- Lectures (board, overhead projector, molecular models, software chemistry programs and data show)
- 4.2- Tutorials and discussion sessions
- 4.3- Practical sessions

5- Student Assessment:

a-Assessment Methods and Weighing:

Quiz to assess the knowledge, understanding and intellectual skills of the course.

Mid-Term written exam to assess the knowledge, understanding and intellectual skills of the previously studied sections of the course.

Practical exam to assess the practical and professional skills gained by the students.

Final written exam to assess all the knowledge and understanding of the different sections of the course.

Oral examination to assess all the intellectual skills and knowledge of the different sections of the course.

1) Class participation: 10%

2) Practical exam:25%

3) Oral exam: 15%

4) Final exam: 50%.

b-Assessment Schedule:

Class participation

Quiz 1: Week 4-5

Quiz 2: Week 8-9

Other activities: throughout the semeste

Practical exam:

Week 13-14

Oral exam:

According to semester timetable

Final exam:

According to semester timetable

6-list of references

Course Notes	On FIS
Required Books	On ELS T.W.Graham Solomons, Organic chemistry, 11th ed., John Wiley and Sons, INC, New York, 2013. Furniss, Hanna Ford, Smith, Tutchell, Vogel's textbook of Practical Organic Chemistry, 5th ed, Longman Group UK Ltd, England, 1989. Finar Organic Chemistry 6th Edition published by Pearson Education (Singapore) Pte. Ltd, Indian Branch; printed in India by A.P. offset Pnt. Ltd. Organic Chemistry 4th Edition by Janice Gorzynski Smith, 2013
Recommended Books	Jerry March, Advanced Organic Chemistry, Reactions, Mechanisms.
Periodicals Web Sites	Different pharmacopoeias (Egyptian & british) J. Org. Chem., J. Med. Chem., Tetrahedron letters, J. Am. Chem.
	Soc., Khan Academy and https://www.educator.com//chemistry/organic-chemistry

Course Coordinator:

Dr/Mohammed Ibrahim Abdel Latif Hamed

Dr Mohammed Mahmoud El_Agawan

-Head of Department : Prof.Dr/ Mona Hetta

Date: •4/•4/Y•15





Physical and inorganic Chemistry

PC 101







Name of Department: Pharmaceutical Analytical Chemistry

A. Basic Information

Program(s) on which the course is given:

Bachelor of pharmacy

Department offering the course:

Pharmaceutical Analytical Chemistry department

Faculty offering the program.

Faculty of pharmacy, Fayoum University

Dept. responsible for teaching the course.

Pharmaceutical Analytical Chemistry department

Academic year / level:

first level, first semester

Course title:

Physical and inorganic Chemistry

Course code.

PC 101

Contact hours (credit hours):

Lecture 2 (2) + Practical 2 (1): Total 4 (3)

Pre-requisite of the course:

Registration

Course coordinator:

Major or Minor element of program:

Major

Date of specification approval:

09 / 09 /2015

B. Professional Information

1. Overall Aims of Course

- 1. Understand the fundamental concepts of thermodynamics, chemical equilibrium, and kinetics.
- 2. Apply the principles of gas laws, solutions, and colligative properties to pharmaceutical systems.
- 3. Analyze the thermodynamic properties of chemical reactions and predict spontaneity.
- 4. Interpret and solve problems related to chemical kinetics and reaction mechanisms.
- 5. Develop proficiency in understanding the behavior of acids, bases, and buffers in pharmaceutical applications.
- 6. Apply the concepts of electrochemistry to processes such as corrosion, redox reactions, and electrolysis.

21





- 7. Evaluate molecular structure, bonding theories, and periodic trends and their impact on chemical behavior.
- 8. Utilize theoretical knowledge to explain physical properties of matter and their application in drug formulation and stability.

2. Intended Learning Outcomes of Course (ILOs)

After successful finishing the course, students should be able to

a- Knowledge and Understanding:

By the end of the course, the students should be able to

a₁- Understand fundamental concepts of physical and inorganic chemistry and their applications in pharmaceutical sciences.

 a_2 - Understand chemical kinetics, thermodynamics, and electrochemistry and how they relate to drug formulation and stability:

b- Intellectual Skills:

b_l- Critically evaluate the physical properties of drugs and lnorganic compounds for pharmaceutical use:

b₂- Solve problems related to chemical reactions, energy changes, and drug interactions using theoretical concepts.

c- Professional and practical Skills:

c₁- Perform laboratory experiments to synthesize and analyze inorganic compounds used in pharmacy.

c₂- Apply principles of physical chemistry in practical pharmaceutical settings, such as drug formulation and stability testing.

d- General and Transferable Skills:

d₁- Collaborate efficiently in a team to solve chemistry-related problems in pharmaceutical contexts.

d₂- Demonstrate time management skills, especially in conducting experiments and managing lab work.

d₃- Apply critical thinking and independent learning to enhance understanding and skills in chemistry relevant to pharmacy practice.





2. Contents

Teaching	Contents	No. of	No. of	
week	TOPIC	lecture	practical	Assessment of ILOs
Week		hours	hours	
	Introduction to Physical and General	Hours		
1	Chemistry: Importance and Applications in	2		a2, d1, c2, b2
_	Pharmacy	-		42, 41, 02, 02
	Lab safety		2	
	Gaseous State: Gas Laws, Kinetic Molecular			1 0 1 4 4 1
2	Theory, and Ideal vs. Real Gases		aniutulos	a1, a2, b1, c1, d2
:	Measurement of Gas Laws: Boyle's Law and	et na	2	
·	Charles' Law		2	
	Thermodynamics I: Basic Concepts of	2		01 02 51 01 42
·., · 3	Energy, Work, and Heat	2		a1, a2, b1, c1, d2
i	Determination of Enthalpy Changes in		2	
	Chemical Reactions (Calorimetry)		Z ,, ,,	
•	Thermodynamics II: Laws of			
4	Thermodynamics and Their Pharmaceutical	2	allive	a1, b1, c1 d1, d2, b2
	Applications	Entransis Control All States	affiliaris diff.	
	Measurement of Equilibrium Constant for a		2	
	Chemical Reaction			
	Chemical Equilibrium: Principles and	2		-1 1 1 1 10 10
5	Applications in Drug Reactions	- 2		a1, b1, c1, d2, d3
	Preparation of Solutions: Molarity, Molality,		3145	
	and Percent Solutions		2	The second secon
	Solutions and Their Properties: Concentration			
6	Solubility, and Colligative Properties	2		a1, a2, b1, c1, d1
_	Determination of Colligative Properties:			
	Freezing Point Depression		2	
	Molecular Structure and Bonding: Lewis			
7	Structures, VSEPR, and Hybridization	2		a1, a2, b1, c1, c3, d2
7	Reaction Rate Studies: Determining the Effect	ie iš	2	
	of Concentration on Rate		Z Î	
	Electrochemistry: Galvanic Cells, Corrosion,	2	· Salah Salah Salah	al a2 a1 a2 d1
8	and Pharmaceutical Relevance	<u> </u>	A STATE OF THE STA	a1, a2, c1, c2, d1
	Construction of a Galvanic Cell and		7重量	
	Measurement of Electrode Potentials		-	
	Chemical Kinetics I: Rate Laws and Factors	7.2	Dept.	a2, b1, c1, d1, b2
9	Affecting Reaction Rates	1.	Contract of the State of the St	
	Investigating the Effect of Temperature on		iii \25.\-r	
	Reaction Rate (Arrhenius Equation) Chemical Kinetics II: Reaction Mechanisms			
10	and Catalysis in Pharmaceutical Reactions	\ \2:		a1, b1, c1, d2
70	Determination of Molecular Structure Using	And the same of	35. 1 6 3	
	VSEPR Models	And the state of t	2	
11	Intermolecular Forces: Impact on Drug	2	MA .	o2 h1 a1 d2 a2
.	orocalar rorces, impact on Drug	. 🗸 📆	1/2	a2, b1, c1, d3, c2





Course Specifications (2015 –2016)

	(=020 -020/		
Stability and Properties			
Final practical exam			
	22	20	
	FINAL Exan	<u> </u>	

Teaching and Learning Methods

- 1- Lectures (board, data show, power point)
- 2- Interactive learning (Discussions, brain storming)
- 3- Self-study (Tutorials)
- 4- Practical (labs, tools, chemicals, glassware, equipment and instruments)
- 5- Other methods (Assignments)

Student Assessment Methods

- 1- Mid erm exam to assess ILO's a1, a2
- 2- Practical exam to assess ILO cland d2
- 3- End-term exam to assess ILO's a, c
- 4- Oral exam to assess ILO's al. c

ssessment Schedule

uiz 1 4th or 5th week

uiz 2 8th or 9th week

ractical exam 11th week

inal exam 12th week

ral exam

Veighting of Assessments

eriodical

ractical

inal exam

Oral exam

100%

Total





Course Specifications (2015 –2016)

6. List of References

- 1. Essential Books (Text Books)
- 2. Principles of Chemistry: A Molecular Approach by Nivaldo J. Tro.
- 3. General Chemistry: The Essential Concepts by Raymond Chang and Jason Over
- 4. Physical Chemistry by Peter Atkins and Julio de Paula
- 5. Quantitative Chemical Analysis by Daniel C. Harris
- 6. Chemistry: The Central Science by Theodore L. Brown
- 7. H. Eugene LeMay, Bruce E. Bursten, and Catherine J. Murphy
- 8. Journal of Chemical Education
- 9. Analytical Chemistry
- 10. Royal Society of Chemistry (RSC) Publications

7. Facilities required for teaching and learning

- 1. Classrooms.
- 2. Laboratory facilities.
- 3. white board, smart board, Data show.
- 4. Library.
- 5. Computers.
- 6. Online educational platforms for teaching, discussing research projects and E-exams (Google classrooms, Google drive and Microsoft Teams).

Course Coordinator: Dr. Abobakt Abdel Wahab

Head of Department: Prof./ Mona Hetta

Date: 09/09/2015

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وحدة ضمان الحودة

Course Specification of Pharmacognosy1

niversity: Fayoum culty: Pharmacy

- Course specifications:

ogram (s) on which the course is given: Bachelor of pharmacy (Clinical

urance Unit

armacy)

partment offering the course: Pharmacognosy

ademic year Level: First level /second term

tte of specification approval: 09/ 2015

Basic information:

tle: Pharmacognosy I

actical: 1 hrs/ week

edit hours: 3 (2+1)

ctures: 2 hrs/ week

tal: 3 hrs/ week

Professional information:

Overall Aims of the Course:

- completion of the course, the student will be able to:
- Describe morphological, histological characters and uses of medicinal flowers, barks, wood and herbs as well as identification of different active constituents and adulteration.
- Examine and determine the active constituents of the studied drugs

code: PG.202

مر وجمة الحدودة



حامعــــة الفيوم كلية الصيدلــــة «حدة ضمان الجودة

2-Intended Learning Outcomes (ILOs):

A	- Knowledge and Understanding
-	
	Illustrate morphological, Histological characters and uses of medicinal flowers,
a	barks, wood and herbs
-	
a2	Identify adulteration of different medicinal flowers, barks, wood and herbs.
-	
$\begin{vmatrix} a \\ a \end{vmatrix}$	Identify different active constituents of medicinal flowers, barks, wood and
4.	herbs.
\mathbb{R}	Professional and Prosting L. I.
<u> </u>	Rentessional and Cractical skills s:
b1	Handel and dispose chemicals in a safe way
b2	Handle lab microscopes effectively
b3	Examine drugs of plant origin in entire and powdered form.
ļ	
<u>C</u> -	Intellectual skills
c1	Adapt GLP and safety guidelines in the lab.
CI	and safety guidennes in the lab.
-	Differentiate hat and 1
c2	Differentiate between drugs in entire and powdered form
	Dicc
c3	Differentiate the active constituents of different drugs.
D- (General and Transferable skills
	Work as a member of a team
dI	Work as a member of a team
**	Implement writing and presentation skills
d2	The state of the sentation skills
	Develop critical thinking dealers to the second sec
d3	Develop critical thinking, decision making and time management skills
<u></u> .	



D- Contents:

Week No.	Lecture contents (2hrs/lecture)	Practical session (1hrs/lab)
1	 General introduction for what will be taught all over the term Introduction for the flowers and giving the students the possible references, web sites, text books. 	Laboratory safety measuresDealing with microscope.
2	 Rest of introduction of flower. Description of clove flower Including macro- and micro-morphological study for entire drug and for powdered clove. 	 Description of clove flower Including macro- and micro- morphological study for entire drug. Histology of powdered clove.
3	Chamomile (German and Roman) and calendula flowers macro-and, micro-morphology of the entire and powdered drugs, chemical identification.	German chamomile flower macro-, and micro-morphology, powders and chemical identification.
4	Rest of flower drugs, uses, active constituents, pharmaceutical preparations.	 Santonica flowers macroand, micro-morphology, powder and chemical identification. Activity (Net research on the pharmaceutical products derived from the flowers).
5	General introduction of bark	Cinnamon bark macro-and micro-morphology, powder and chemical identification.
6	Rest of the introduction and cinnamon bark: macro-and; micro-morphology-, powder and chemical identification.	Cassia barks (Morphology, histology, powder and chemical test, when it is possible.
7	 Cinchona in details: macro-and; micro-morphology, powder and chemical identification. Periodic exam. 	 Cinchona barks (Morphology, histology for entire drug powder and chemical test when it is possible. Morphological demonstration for all barks.

acy											
e	Unit										
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(Morphology, histology, powder and chemical test when it is possible. Cascara and frangula in details: macromorphology; micro-morphology powder and	 Revision for powdered barks (Cinnamon, Cassia and cinchona). Quassia wood: Macro- and micro-morphology of entire wood. Galls: Macro-morphology and micro-morphology of entire drug.
• Rest of the taught barks.	Hyoscymus muticus herb: Macro- and micro-morphology of entire drug
 Introduction to wood Galls: Macro-morphology of entire wood; micro-morphology-, Powder; chemical identification. 	Mentha: Macro- and micro- morphology of Powder and chemical identification
Herbs introduction Herbs introduction Official and nonofficial herbs	 Practical exam 1 morphology of entire woods Practical exam 2
• Revision.	
• Final written exam.	

13.3 1.3

Unit	
Cascarilla, quillaia, and canella bark (Morphology, histology, powder and chemical test when it is possible.	Revision for powdered barks (Cinnamon, Cassia and cinchona).
Cascara and frangula in details: macro- morphology; micro-morphology powder and chemical identification.	micro-morphology of entire drug.
Rest of the taught barks.	Hyoscymus muticus herb: Macro- and micro-morphology of entire drug
 Introduction to wood Galls: Macro-morphology of entire wood; micro-morphology-, Powder; chemical identification. 	Mentha: Macro- and micro- morphology of Powder and chemical identification
 Herbs introduction mornhology micro-morphology powder ar Official and nonofficial herbs 	Practical exam 1 morphology of entire woods Practical exam 2
• Revision.	
• Final written exam.	



جمع به العيوم كلية الصيدل ة وحدة ضمان الجودة

E- Teaching and Learning Methods:

- Lectures and interactive lectures.
- Practical sessions
- Self learning (group discussion, group assignment)

F- Student Assessment methods:

- Periodic exam to assess: a1, a2, a3, c1,c2 and c3
- Written exams to assess: a1, a2, a3, c1, c2 and c3
- Practical exams to assess: b1, b2, b3 and d1
- Oral exam to assess: a1, a2, a3, c1, c2 and c3
- Activities (group discussion, group assignment, case study) to assess: d1,
 d2, d3





Assessment schedule

Assessment (1): activity	Week 4	
Assessment (2): Periodic exam	Week 7	
Assessment (3): Practical exam	Week 12, 13	
Assessment (4):Final written exam	Week 15	, '
Assessment (5): Oral exams	Week 15	
(-): (-):		<u>.</u>

Weighting of Assessment

Assessment method	Marks	Percentage
Periodic exam	10	10%
Practical exam	25	25%
Final written exam	50	50%
Oral exam	15	15%
TOTAL	100	100%

G-Facilities required for teaching and learning:

- For lectures: Black (white) boards and data show.
- For Labs: Chemicals, glassware and microscopes.

There are large to the second of the second

H- List of References:

1- Course Notes: Student book of Pharmacognosy I approved by Pharmacognosy Department (2019).

2- Essential Books:

- Wallis, T. (1967). "Text Book of Pharmacognosy 5th Edition, London." <u>J and A.</u> Churchhill Ltd: 81-82.
- De Smet, P. A., K. Keller, R. Hänsel and R. F. Chandler (1992). <u>Adverse effects of herbal drugs</u>, Springer.
- Betty, P. and W. Derek (2000). "Atlas of microscopy of medicinal plants, culinary Herbs and species." <u>CBS publisher New Delhi</u>: 17-42.



- Evans, W. (2005). Trease and Evans Pharmacognosy, ed: 15th, Elsevier, a reed Elsevier India pvt ltd.
- Jackson, M. and A. Lowey (2010). <u>Handbook of extemporaneous preparation: a guide to pharmaceutical compounding</u>, Pharmaceutical Press London, UK.
- Upton, R., A. Graff, G. Jolliffe, R. Länger and E. Williamson (2016). <u>American herbal</u> <u>pharmacopoeia: botanical pharmacognosy-microscopic characterization of botanical medicines</u>, CRC Press.
- McCreath, S. B. and R. Delgoda (2017). <u>Pharmacognosy: Fundamentals, applications and strategies</u>, Academic Press.

3- Recommended Books

- Martindale (2007), "The extra pharmacopeia". 31st edn., by James, E.F Reynolds. And Kathleen Parfitt, Royal Pharmaceutical Society, London.

4- Periodicals and websites:

- Aquilina A. (2013), The extemporaneous compounding of paediatric medicines at Mater Dei Hospital. <u>Journal of the Malta College of Pharmacy Practice</u>. Issue 19, 28 30.
- http://canadianpharmacistsletter.therapeuticresearch.com/ce/ceCourse.asp
- https://www.google.com/search?safe=active&sxsrf=ACYBGNT1wfCQl6DGxZ5ouZYl1QZZfJSrYg:1568843605556&q=Pharmacognosy4all&tbm=isch&source=univ&sa=X&ved=2ahUKEwiel8TurdvkAhVIrxoKHcTHDMAQ7Al6BAgBECQ&biw=1008&bih=584#imgrc=7NmuWomEPl70WM:

Course Coordinators: Prof. Dr. Mona Hetta

Head of Department: Prof. Dr. Mona Hetta

Date:



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ILOs	professional and practical skills	b2)	ĺ	家 [W) i										
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	knowledge and understanding	a2				×							×					×				×	
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	Course Contents		Lectures		General introduction for what will	be taught all over the term	 Introduction for the flowers and 	giving the students the possible	references, web sites, text books.		 Rest of introduction of flower. 	 Description of clove flower 	Including Macro- and micro-	morphological study for entire	drug and for powdered clove.	• Chamomile (German and Roman)	and calendula flowers macro-and,	Micro-morphology of the entire	and powdered drugs, chemical	identification.	• Rest of flower drugs-, uses-, active	constituents, pharmaceutical	preparations.
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General introduction of bark.

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	Rest of the introduction and cinnamon bark: macro-and; micro morphology -, powder and chamical identification	Cinchona in details: macro-and; micro-morphology - powder and chemical identification.	• Cascarilla, quillaia, and canella bark (Morphology, histology, powder and chemical test when it is possible.	Cascara and frangula in details: macro-morphology; micro- morphology powder and chemical identification.	• Rest of the taught barks.	Introduction to wood Galls: Macro-morphology of entire wood; micro-morphology-, Powder; chemical identification.	Herbs introduction Official and nonofficial herbs	Practical sessions	 Laboratory safety measures Dealing with microscope. 	- Doggan man grands
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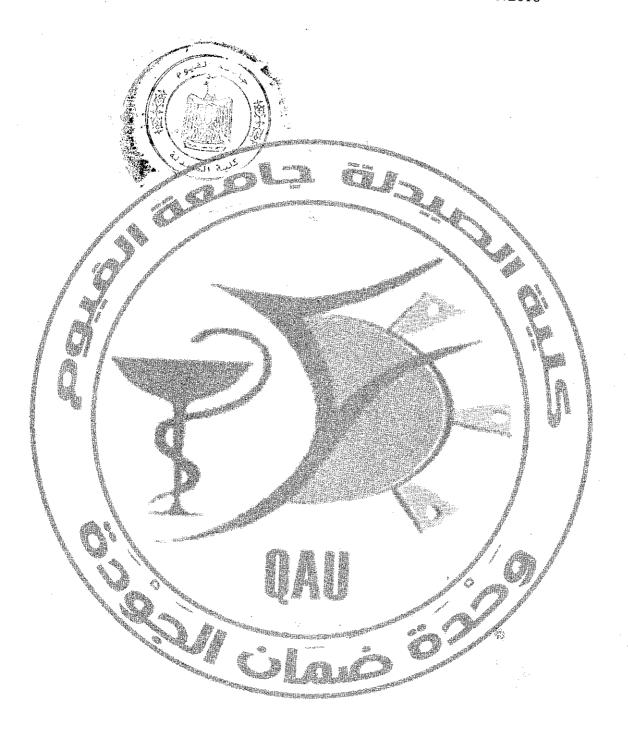


Course Specifications
(2018 – 2019)

Head of Department: Ass. Prof. Mona El Naa

Date:

/09/2018



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morphological study for entire	Including Macro- and micro-	 Description of clove flower 			urance Unit









Course Specifications (2017 -201°)

Pharmaceutical Analytical Chemistry I PC 101





Course Specifications (2017 - 201°)

Pharmaceutical Analytical Chemistry

asic Information

ogram(s) on which the course is given

epartment offering the course

cademic year / level

ourse title

ourse code

ontact hours (credit hours)

culty offering the program

e-requisite of the course:

ourse coordinator

ajor or Minor element of program

ate of specification approval

Bachelor of clinical pharmacy

Pharmaceutical Analytical Chemistry department

Faculty of pharmacy, Fayoum University

ept. responsible for teaching the course Pharmaceutical Analytical Chemistry department

First level, first semester

Pharmaceutical Analytical Chemistry I

PC 101

Lecture 2 (2) + Practical 2 (1) Fotal 4 (3)

Registration

Dr. Hany Mohamed Gamal

Major

09/09/2014

<u>rofessional Information</u>

verall Aims of Course

The aim of the course is to provide students with the theoretical basis and practical aspects of the different methods of different quantitative (acid-base, preciptimetric) analysis.

Develop the practical skills in analyzing chemical compounds that enable the graduate to communicate efficiently and effectively with health care professionals.

Use educational methods that make graduates be life-long learner, creative researcher and effective participant in healthcare of the community.

Acquire the ability of integration, information evaluation and solving diverse analytical problems in an efficient way.

Intended Learning Outcomes of Course (ILOs)

Knowledge and Understanding:

y the end of the course, the students should be able to

- 1- Explain the basic principles of quantification of different chemicals
- 12- Acquire much information about different quantitative analytical methods.
- 3- Recognize the suitable analytical methods for identification of different harmaceuticals
- 4- Understand the basics of some physico chemical specifications as pH, pK_a, K_{sp} of 'arious materials used in pharmaceutical preparation.

وعدة حمان الحية







Course Specifications (2017 -2019)

- a₅- Recall the basic principles of pharmaceutical science including calculations of concentration, writing chemical formulas, determination of percent purity.
- a₆- Define the basic principles of chemical reactions, chemical equilibria and balance chemical equations.
- a₇. Recall the basic principles of quanitative (acid-base, preciptimetry) chemical analysis of cations and anions.

b- Intellectual Skills

By the end of this course, the student should be able to:

- b₁- . Select appropriate analytical methods required for conformity of specifications of raw material.
- b₂. Assess different methods for quanitative chemical analysis of different substance.
- b_{3-} . Describe an analytical scheme for analysis of cations and anions.
- b₄ Write a report on analysis of a chemical substance.
- b₅- Demonstrate the principles and limitations of practical techniques.

c- Professional and Practical Skills

By the end of the course, the student should be able to:

- c₁. Apply the safe handling and safe disposal of chemicals according to the ethical and legal guidelines.
- c₂. Apply the practical methods required for quantification of different compounds.
- c₃- Standardize chemical reagents used in acid-base quantitative analysis
- c4- Analyze different pharmaceutical substances through acid-base, & preciptimetric analysis.

d-General and Transferable Skills

By the end of the course, the student should be able to:

- d_1 . Interact and communicate by verbal and written means with other health care professionals in their own specialized language.
- d₂. Team-working in diverse pharmaceutical & social settings.
- d₃. Keep up with the pharmacy profession and pharmaceutical industry as a life -long independent continuing education post-graduation







Course Specifications (2017 -201°)

 d_4 . Apply proper safety measures according to standard guidelines

d₅- work efficiently in laboratory.

 d_6 . Analyze, evaluate information and solve problems.

3. Contents

3. Cu	ntents	· · · · · · · · · · · · · · · · · · ·		
Teaching week	TOPIC	No. of lecture	No. of practical	Assessment of ILOs
		hours	hours	
1	Introduction, Rate of the reaction	2	State of the state	a2, b1, d1
	Lab safety		100m 2 to 100	
2	Chemical equilibrium	2		a1, a2, a3, b1, b2, c1, c2, d2
	Standerdization		2	
3	Chemical Kinetics	2		al, a2,a4,b1,b2, c1, c2, d2
	Assay of HCl	156152150 290	2	
4	Introduction to qualitative & quantitative analysis			a1, a5,a3,b1,b3, c1, c3, d1,d2
	Assay of HCl/Acetic mixture	100	2	
5	Acid-base theory	100 2 100 100 100 100 100 100 100 100 100 100	ors resolution of den de seccionistica año conferencia	a1, a2,a4,b1,b2, c1, c3, d2, d3
	Assay of formaldyde		2	
6	Acid-base titration curve & buffer. Quiz 1	2		a1, a2,a3,b1,b3, c1, c2, d1,d5
	Assay of KOH	Service of	2	
~	Application acid-base titrations.	2		a1, a2, a4, b1, b2, c1, c3, d2, d6
7	Assay of salts	n was sam	2	
8	Application acid-base titrations.	2		a1, a2, a3, b2, c1, c2, d1
	Ksp problems I	Sid, follows:	2	
9	Preciptimetric methods.	2	A STATE OF THE STA	a2,a3,b1, c1, c2, d1,d6
	Ksp problems II		2	
10	Ksp problems 11 Preciptimetric methods. Quiz 2	2	All the second s	a1, a4,a4, a5, a6, a7 b1,b2, c1, c3, c4 d2
	Revision	43.6489880xxxxx	2	,
11	Preciptimetric methods.	2		a2,a4, a5, a6, a7 b1, c1, c2, c4, d1, d5, d6
	Final practical exam	· iii	1 37 200	The state of the s
Total no	T	22	// 20	a Maria
of hours			<u> </u>	
11 & 12	F)	INAL Exa	m	
		À	· · · · · · · · · · · · · · · · · · ·	







Course Specifications (2017 -201°)

eaching and Learning Methods

l- Lectures (board, data show, power point)	\checkmark	
2- Interactive learning (Discussions, brain storming)	1	
3- Self-study (Tutorials)	-	
t Dragtical (labor tools abomicals alaceware-againment and instruments)	J	

5- Other methods (Assignments)

tudent Assessment Methods

- 1. Written exams to assess knowledge and understanding as well as intellectual skills
- 2. Oral exams to assess all types of skills and mainly general and transferrable skills actice.
- 3. Practical exams

ssessment Schedule

aiz 1		ilinus exercisada = 1995 il julyar 1881 — 1981	and the second	th week	
ıiz 2				AND REPORTS	
actical exa	m		477	l ^{ih} week l2 th week	
nal exam	A Alba.			12 week 12 th week	
70	T OF Arga	Samonts		12 Week	
leighting	g of Asse:	22 ments		15	%

Total				100%
ral exam	 	- 100 Maria	Переноват (ST-14	10%
	en state of the st			
nal exam	WAT	他	All and the	50%
actical	A)	reservations and the second	lintona, tolkie	25%

ist of References

1- Course Notes

omplied by the department







Course Specifications (2017 -201°)

6.2- Essential Books (Textbooks)

- J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas "Vogel's Textbook of quantitative chemical analysis" by Pearson education (6th edition) 2004
- R.de Levie "principles of quantitative chemical analysis" by Mc Graw-HillCompanies.inc.1997.
- G. D. Christin "Analytical chemistry" by John Wiley & sons inc. (5th edition) 1994

6.3- Periodicals

- -Analytical letters
- -J. pharmaceutical and biomedical analysis
- -Analytical chemistry

6.4- Web Sites

https://www.degruyter.com/view/journals/revae/revac-overview.xml

7. Facilities required for teaching and learning

- 1. Lecture rooms with data show
- 2. Procurement of latest edition of the above-mentioned texts and others to update the education process

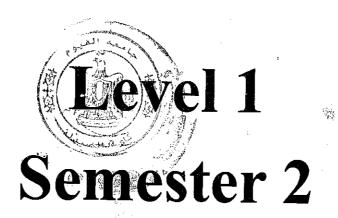
Course Coordinator: Dr. Hany Mohamed Gamal

Head of Department: Pr. Mona Hetta

Date: 09 /09/201°

Course Specifications

Physical pharmacy (PT201)





Clinical Pharmacy Program



Course Specification

A-Basic Information	
Course code:	PT201
Course name:	Physical pharmacy
Credit hours of the course:	Lecture: 2 Practical: 1 Total: 3
Pre-requisite of the course:	Registration
Department teaching the course:	Department of Pharmaceutics
Program for which the course is given:	Clinical Pharmacy Program
Course Co-ordinator:	Dr.Eman Ahmed Mazyed
Head of the Department:	Prof Dr. Mona Hetta
Date of specifications approval:	09/09/2014

B-Professional Information

1- Overall aims of the course:

At the end of this course the student must be able to understand:
Important pharmacy-related physical principles in areas such as states of matter ,phase equilibrium and phase rule , solutions and disperse systems , adsorption ,solubility ,dissolution rate , buffers , isotonic solutions and rheology .

2-Intended learning outcomes (ILOs):

a-Knowledge and Understanding:

By the end of this course, the student should be able :

Spence

- A1. Define basic principles and different terms of physical pharmacy..
- A2.Understand physical principles of states of matter and phase rule
- A3. Recognize basic understanding of physical properties of solutions, buffers.
- Isotoninicity, disperse systems and Rheology..
- a4.. Understand physicochemical properties of drugs including solubility, distribution and adsorption. **b-Intellectual Skills**:
- By the end of this course, the student should be able to
- bl Apply information regarding physical principles in deigning dosage forms.
- b2 Solve problems related to states of matter, concentration expression, buffers and isotonic solutions
- b3 Recognize basic rules and equations regarding physical principles essential for pharmaceutical applications
- .Differentiate between different types of solutions, buffers and surfactants.b4.
- .B5.Compare between different types of flow and rheograms
- B6.Select suitable calculations necessary for pharmaceutical experimental procedures..

c-Professional and Practical Skills:

By the end of this course, the student should be able to

- cl. Use suitable method for calculating solubility, viscosity and surface tension.
- C2. Operate different pharmaceutical laboratory instruments used in determining various physical properties such as surface tension, viscosity, adsorption and solubility..

d-General Skills:

By the end of this course, the student should be able to::

- d1. Use computer and modern technologies.
- d2. Gain the skills of self- learning and presentation.
- d3 Work effectively in a team to measure and understand various physical properties of drug solutions.
- d4. Calculate and properly express solution concentrations:
- D5.Demonstrate critical thinking, problem solving and decision making abilities in a variety of heoretical and practical situations in pharmacy practice.
- 16.Prepare and discuss scientific reports.



Clinical Pharmacy Program

Buffer and buffer capacity.	2+1	2	1	
Revision	2+1	_	1	
Revision	2+1		1	
	39	26	13	
	buffer capacity.	buffer capacity. Revision 2+1	Revision 2+1 2 Revision 2+1 2	Revision 2+1 2 1 Revision 2+1 2 1

4- Teaching and Learning Methods (lectures, open discussion, role plays,...etc):

- Lectures, PowerPoint presentation
- Tutorial classes
- Open discussion
- Research assignment and interactive discussions.

5- Student Assessment:

a- Assessment Methods and Weighing:

- Class participation: 10%
- Practical Exam: 25%
- Oral Exam: 15%
- Final Exam: 75%

b- Assessment Schedule:

- Class participation: Quiz 1: Week 4
 - Quiz 2: Week 8
 - Other activities: throughout the semester
- Practical Exam:
- Week 13
- Oral Exam: Final Exam:
- According to semester timetable According to semester timetable



6- List of References:	
Course Notes	Course notes, Hand outs
Required Books	Physical Pharmacy, David Attwood and Alexander T. Florence (editors), Pharmaceutical Press, London, UK, 2008
	Agarwal, S.P., and R. Khanna. Physical Pharmacy 1st ed. CBS Publishers and Distributers, 2002.
	-Alton, Michael E. Pharmaceutics: The Science of Dosage Form Design. 2 nd ed Churchill, 2002
Recommended Books	-Cherng-Ju Kim, Advanced pharmaceutics "Physicochemical
	Principles" CRC press, 2000.
	-University of the Sciences in Philadelphia, ed. Remington. The Science and Practice of Pharmacy. 21 st ed. Lippincott, 2005.
	-Alfred N. Martin Physical Pharmacy 4th ed. Lea & Fibiger, 1996.
	J. Parma Pharmacology, Ind. Pharm, Pharmazie, Drug Devel. Ind. Pharm,
	E-book,
Periodicals	http://www.nordicrheologysociety.org
Websites	http://www.rpi.edu/dept/chem-
	eng/Biotech Environ/Adsorb/adsorb,htm

Course Coordinator: Dr. Eman Ahmed Mazyed

Head of Department: Prof. Mona Hetta

Websites ---

Course Coordinator: Dr. Eman Ahmed Mazyed

Head of Department: Prof. Dr. Mona Hetta

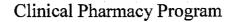
Date: 09/09/2014

Course Specifications

Pharmacy Orientation (PT 202)

Level 1
Semester 2







Course Specification

A-Basic Information	
Course code:	PT202
Course name:	Pharmacy Orientation
Credit hours of the course:	Lecture: 2 Practical: Total: 2
Pre-requisite of the course:	Registration
Department teaching the course:	Department of Pharmaceutics
Program for which the course is given:	Clinical Pharmacy Program
Course Co-ordinator:	Dr/Eman Ahmed Mazyed
Head of the Department:	Dean of the Faculty of pharmacy
Date of specifications approval:	09-09-2014

B-Professional Information

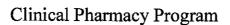
1- Overall aims of the course:

At the end of this course the student should be able to understand:

- o The importance of pharmacy and the role of pharmacist.
- o Different routes of drug administrations and different dosage forms.
- o Medical terminology and pharmaceutical terminology.
- o Rational use of drugs.
- o Ethics of pharmacy, Self-care and Self-medications.
- o The history of pharmacy and a focus about Pharmacy in the Pharaonic age, role of Arabs in pharmacy.

Meno







2- Intended learning outcomes (ILO's):

a-Knowledge and Understanding:

By the end of this course the student should be able to:

- al. Define pharmacy, drug and classification of drugs.
- a2. Recognize pharmacy profession, prescription parts, difference between types of pharmaceutical dosage forms.
- a3. Describe the scope of patient services, drug services, and the prescription.
- A4. Understand various roles & ethics of pharmacy.
- a5. Plain different types of pharmaceutical dosage forms, the routes of drug and factor affecting them.
- a6. know the history of pharmacy.

b-Intellectual Skills:

By the end of this course, the student should be able to:

- b1. Distinguish between the units of weight, volume and length.
- b2. Apply pharmaceutical knowledge in the formulation of safe and effective medicines as well as in dealing with new drug delivery systems.
- b3. Apply pharmaceutical knowledge about prescriptions and how to prepare them in proper way.
- b4. Control possible physical and/ or chemical incompatibilities that may occur during drug dispensing.

c-Professional and Practical Skills:

By the end of this course, the student should be able to:

- c1. Use the proper pharmaceutical and medical terms, abbreviations and symbols.
- c2. Advise patient and other health care professionals about safe and proper use of medicines.
- c3. Calculate the dose for children, weight, volume and length.

d-General Skills:

By the end of this course, the student should be able to:

- dl. Use computer and modern technologies
- d2. Gain the skills of self-learning and presentation.



d3. Work in group.



Clinical Pharmacy Program

3- Course contents:					
Topic		No. of hours			
	Lecture	Practical	Total		
History of Pharrnacy(in Arabic)	2		2		
History of pharmacy (cont.)	2		2		
History of pharmacy (cont.)	2		2		
Scope of pharmacy, the functions of pharmacist and Drug information resources	2		2		
Routes of drug administrations	2		2		
Dosage forms	2		2		
Examples of medical terminology and pharmacy terminology	2		2		
Rational use of drugs, Pharmaceutical care, self-care and self-medications	2		2		
Ethics in pharmacy	2 .		2		
Standard of practice	2		2		
Prescriptions	2	- <u>-</u>	2		
Different types of incompatibilities, and how can we solve them	2		2		
Revision	2		2		
Fotal	26		26		

Course specification

Course code:	PC 203
Course name:	Pharmaceutical Organic
	Chemistry II
Credit hours of the course:	Lecture: 2
	Practical: 1
	i racticai. 1
	Total: 3
Pre-requisite of the course:	Pharmaceutical Organic
	Chemistry I
Department teaching the	Pharmaceutical Medicinal
course:	Chemistry Department
Program for which the course	Clinical Pharmacy Program
is given:	
Course Co-ordinator:	Dr Mohammed Ibrahim
	Abdel_Latif Hamed.
	Dr/ Mohammed Mahmoud
	El_Agawany
Head of the Department:	Prof. Dr./ Mona Hetta
Date of specifications	09/2015
approval	

June of June

b-professional information

1-Overall aims of the course:

The aim of this course to ensure that graduates have the necessary knowledge, understanding and skills to apply the basic chemistry of aliphatic organic compounds.

2-Intended learning outcomes (ILO's):

Knowledge and Understanding:

Intellectual Skills:

By the end of this course, the student should be able to:

al-Know the basic concepts of aromaticity and chemistry of arenes

a2-Know the characteristics of chemistry of benzenoids and polynuclear compounds

a3-Understand The basics of spectroscopy and spectral analysis of organic compounds

By the end of this course, the student should be able to:

- b1- Recognize organic compounds from their physical and chemical properties.
- b2- Discuss organic problems of different organic classes.
- b3- Apply suitable methods for synthesis, purification, and identification of chemical substances from different origins.
- 64- Analyze and interpret experimental studies as well as data from published literature using spectroscopy.

c- Professional and Practical Skills:

By the end of this course, the student should be able to:

- c1-Select the appropriate methods for identification and synthesis of organic compounds
- c2- Identify aromatic organic compounds and carbohydrates physically and chemically.
- c3- Analyze spectral data of organic compounds.

d-General Skills:

By the end of this course, the student should be able to:

- d1- Use the literature via different information sources including the library, internetetc.
- d2- Care about himself, his colleagues and all the tools and instruments of his lab during dealing with chemical organic compounds

3-course contents

Topics	lecturer	lecture	practical	
Introduction to organic chemistry, structure of benzene and its stability, (resonance), and the aromaticity phenomena. Classification and Nomenclature of aromatic compounds. Preparation and addition reactions of benzene	Dr/Mohammed Ibrahim	2	1	tota 1
Electrophilic substitution reactions of benzene. Orientation (o-p-m-directing group) of substituted benzene.	Dr/Mohammed Ibrahim	2	1	1
Arenes and their derivatives. Aryl halides.	Dr/Mohammed Ibrahim	2	1	1
Aromatic alcohols Aryl ether.	Dr/Mohammed Ibrahim	2	1	1
Nomenclature, Synthesis, preparation and chemical reactions of aromatic aldehyde and ketone.	Dr/Mohammed Ibrahim	2	1	1
Aromatic carboxylic acids and their derivatives. Identification, Synthesis, preparation and	Dr/Mohammed Ibrahim	2	1	1
nitro compounds.	Dr/Mohammed El Agawany	2	1	1
Aromatic diazonium compounds. Polynuclear aromatic hydrocarbons (Naphthalene. Anthracene, phenanthrene, and anthrquinone)	Dr/Mohammed El Agawany	2	1	1
Introduction to spectroscopy, relation between light and organic compounds, Ultraviolet-and and visible U.V. Beer ambert low, type of electronic orientations alculate of Amax for dienes by using Woodward (Fisher rule)	Dr/Mohammed El Agawany	2	1	1

Introduction to mass spectro fragmentation and Mc-Lafferty rearangment Introduction infrared, Hook low, region of different type of fire sti	Agawany2	2	1	1
Introduction to NMR type	Dr/Mohammed El Agawany Dr/Mohammed El	2	1	1
NMR. Chemical Shift and TMS. Chemical shift of different type of protons. "C-NMR chemical shift. Problems on C-NMR)	Agawany	2	1	1
Touch		24	12	36

4-Teaching and Learning Methods (lectures, open discussion, role plays, etc):

- 4.1- Lectures (board, overhead projector, molecular models, software chemistry programs and data
- 4.2- Tutorials and discussion sessions
- 4.3- Practical sessions

5- Student Assessment:

a-Assessment Methods and Weighing:

Quiz to assess the knowledge, understanding and intellectual skills of the course.

Mid-Term written exam to assess the knowledge, understanding and intellectual skills of the previously studied sections of the course.

Practical exam to assess the practical and professional skills gained by the students.

Final written exam to assess all the knowledge and understanding of the different sections of the course.

Oral examination to assess all the intellectual skills and knowledge of the different sections of the

1) Class participation: 10%

2) Practical exam:25%

3) Oral exam: 15%

4) Final exam: 50%.

b-Assessment Schedule:

Class participation

Quiz 1 Week 4-5

Quiz 2: Week 8-9

Other activities: throughout the semester

Practical exam:

Week 13-14

Oral exam:

According to semester timetable

Final exam:

According to semester timetable

6-list of references

Course Notes	On ELS
Required Books	Essential organic chemistry. Paula Yurkanis Bruice
Recommended Books	a. Fundamental of Organic Chemistry, J. McMurry and E. Simanek. 6th Edition Organic Chemistry: R.T. Morrison and R.N. Boyd:
	6 Edition
	Chemistry for Pharmacy Students. Satyajit D. Sarker and LutfunNahar. 2007)
	Mechanisms in Advanced Organic Chemistry. R.P. Narain; (2008)
	b. Stereochemistry: V. M. POTAPOV, (1979)
	c. Organic Stereochemistry, Michael J. T. Robinson (1999)
·	d. VOGEL's "Textbook of practical organic
	chemistry" Brian S. Fumiss, Antony J. Hannaford,
	Peter W. G. Smith and Austin R.
	Tatchell 5 Edition
Periodicals	Different pharmacopoeias (Egyptian & british)
Web Sites	Journal of Organic Chemistry
	http://www.pubmed.com

Course Coordinator:

Dr/Mohammed Ibrahim Abdel Latif Hamed

Dr Mohammed Mahmoud El_Agawany

-Head of Department: Prof.Dr/ Mona Hetta

Date: 09/09/2014







Faculty of Pharmacy

Clinical Pharmacy Program (2015-2016)

Course Specifications

A-Basic Information			
Course code:	MD 203		
Course name:	Histology		
Credit hours of the course:	Lecture: 2		
	Practical:1		
	Total:3		
Pre-requisite of the course:	No		
Department teaching the course:	Faculty of Medicine, Anatomy and Histology Department		
Program for which the course is given:	Clinical Pharmacy Program		
Course Co-ordinator:			
Head of the Department:	Prof. Mona Hetta		
Date of specifications approval:	09/09/2015		

B-Professional Information

1-Overall aims of the course:

The course aims to enhance the student's knowledge about various tissues (epithelial, connective, muscular and nervous), heart, blood vessels, lymphatic organs, skin and its appendages, systems (digestive and associated glands, respiratory, urinary, reproductive, central nervous system), endocrine glands and eye histology.

2-Intended learning outcomes (ILO's):

a- Knowledge and Understanding:

Page 1 of 5

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Faculty of Pharmacy

Clinical Pharmacy Program (2015-2016)

By the end of this course, the student should be able to:

- al- Define histology, types of microscopes and the principles of staining with hematoxylin and eosin
- a2- Describe light microscopic and electron microscopic features and the function of cell organelles, cell inclusions, and the nucleus
- a3- Discuss the general characteristics of epithelium, its types, sites and the structure of each type.
- a4. Describe the general characteristics of connective tissue (CT), types of CT cells, the structure and types of CT fibers, and types and sites of CT proper.
- a5. Describe the structure and function of red blood corpuscles, white blood cells and platelets.
- a6. Discuss the general characteristics of cartilage, types of cartilage cells and the structure and sites of different types of cartilage.
- a7. Describe the general characteristics of bone, types of bone cells, the structure and sites of different types of bone, and types of ossification.
- a8. Describe the different types of muscle (skeletal, cardiac and smooth muscle)
- a9. Describe the neuron, types of nerve fibers & types of ganglia.
- a10. Describe the histological structure and function of Gastrointestinal tract.
- a.11 Describe the histological structure and function of Endocrine system [suprarenal gland, thyroid gland, pituitary gland]
- a.12 Describe the histological structure and function of Respiratory system [trachea, lung].
- a13. Describe the structure and function of each type of lymphatic organs (lymph node, spleen and tonsil).
- a.14 Describe the structure and function of skin

b-Intellectual Skills:

By the end of this course, the student should be able to:

- b1. Analyze the ultrastructural details of the cell & correlate between the predominance of a certain cell organelle & the function of its cell.
- b2. Correlate between the structure & function of every organ according to its type of cells.
- b3. Compare between the different types of tissues according to their structure, function and distribution in the human body.







Faculty of Pharmacy

Clinical Pharmacy Program (2015-2016)

c-Professional and Practical Skills:

By the end of this course, the student should be able to:

- c1. Differentiate between different tissues and organs in histological slides using light microscope.
- c2. Apply different types of stains specific for different types of cells and tissues.
- c3. Identify the ultrastructural details of cells through electron microscopic pictures.

d-General Skills:

By the end of this course, the student should be able to:

- d1- Apply how to work effectively in a team.
- d2- Express himself freely and adequately by improving his descriptive capabilities & enhancing his communication skills
- d3- Maintain professional image in manner, dress speech and interpersonal relationships that is consistent with the medical profession's accepted contemporary standards in the community.

3-Course contents: No of hours **Total** Lecture **Practical** Introduction to Histology 1 3 The cell structure and function [cell membrane] 1 3 mitochondria- SER- RER- Secretory vesicles The cell structure and function [lysosomes-ribosom 1 3 cytoskeleton-nucleus] Epithelium (general characters- surface epithelium-1 3 glandular epithelium. - myoepithelium. neuroepithelium)







Clinical Pharmacy Program (2015-2016)

Connective tissue [general characters- free and fixed CT	2	1	3
cells - fibers and matrix & types of C.T]		1	
Blood (Erythrocytes leukocytes, blood platelets)	2	1	3
Cartilage [General characters of cartilage – structure of cartilage - types and sites of cartilage] and Bone [General characters of bone – structure of bone - types and sites of bone – bone ossification]	2	1	3
Lymphatic organs (structure & function of thymus, lymph nodes, spleen and tonsils)	2	1	3
Muscular tissue [general characters, structural details of skeletal, cardiac, and smooth muscles]	2	1	3
Nervous tissue [structural details of the neuron, types of neurons, myelination, structure and function of neuroglial cells]	2	1	3
Gastrointestinal tract [oral cavity, salivary gland, liver, pancreas] and Skin	2	1	3
Endocrine system [suprarenal gland, thyroid gland, pituitary gland] and Respiratory system [trachea, lung]	2	1	3
Total	24		36

4-Teaching and Learning Methods (lectures, open discussion, role plays, ..etc):

- Lectures.
- Practical sections mainly using light microscope.
- Open discussion.
- Assignments.
- Role plays.

5- Student Assessment:

a-Assessment Methods and Weighi

- Written exams evaluate the levels of knowledge and understanding and Intellectual Skills.
- Periodic exams evaluate the levels knowledge and understanding and Intellectual Skills.
- Practical exams evaluate the level of the practical Skills.
- Class participation: 10. %
- Practical exam: 25 %
- Final exam: 65. %







Faculty of Pharmacy

Clinical Pharmacy Program (2015-2016)

b-Assessment Schedule:

Class participation: Quiz 1: Week 4-5

Quiz 2: Week 8-9

Other activities: throughout the semester

- Practical exam: Week 12

- Final exam:

According to semester timetable

6-List of References:	
Course Notes	 Human Anatomy and Histology department Book Practical book of Histology for medical students' course
Recommended Books	- Histology: the Big Picture by John F. Ash; David A. Morton; Sheryl A. Scott
Web Sites	- https://www.ncbi.com

Course Coordinator:

Head of Department: Prof. Mona/Hetta

Date: 09/09/2015







Course Specifications (2015 – 2016)

Course: Human Rights

Course code: HU201





Course Specifications (2015 – 2016)

The Samuel Fayour Under

asic Information

ogram(s) on which the course is given:

Clinical

culty offering the program

Pharmacy

ept. responsible for teaching the course

Sociology faculty of arts

cademic year/level

Second semester

ourse title

Human rights

ourse code

HU201

ontact hours (credit hours)

2

e-requisite of the course:

Nο

ourse coordinator

ajor or Minor element of program

Minor

ate of specification approval

09/09/2015

<u>rofessional Information</u>

verall Aims of Course

يغطي هذا المقرر حقوق الانسان في القانون الجلائي، حق الانسان في تغيير جنسته او التخلي عن احدي جنسياته، المواثيق الدوا المتعلقة بحماية حقوق الامسان، علاقة العوامنة و التنمية بالحقوق الاقتصادية و الاجتماعية و الثقافية، الحقوق الاقتصادية الاجتماعية و الثقافية للانسان، حقوق الالبنان في الشريعة الاسلامية، حقوق المرا في قانوش العمل و التأمين الاجتماعي، حقو الانسان في التقاضي، الوقوق المدنية و السياسية للانسان

Intended Learning Outcomes of Course (ILOs)

Knowledge and Understanding:

y the end of the course, the students should be able to:

نكر حقوق الانسان في القانون الجنائي ..

تعديد الحقوق الاقتصادية و الاجتماعية و الثقافية !

فهم حقوق الانسان في الشريعة الاسلامية .:

رسومة حمان الحرودة

Intellectual Skills

مديركس

A CRI







Course Specifications (2015 - 2016)

مناقشة حقوق الإنسان في القانون الجنائي. b.1

. مناقشة حقوق المرأة في قانوني العمل و التأمين الاجتماعي b2

By the end of this course, the student should be able to:

مناقشة علاقة العولمة و التنمية بالحقوق الاقتصادية و الاجتماعية و الثقافية، الحقوق الاقتصادية و الاجتماعية و . 63 الثقافية للانسان

c- Professional and Practical Skills

ق مبادئ حقوق النسبان المتعلقة بالقانون الجنائي .c1

تطبيق حقوق المرأة في قانوني العمل و القامين الاجتماعي .02

By the end of the course, the student should be able to:

d- General and Transferable Skills.

By the end of the course, the student should be able to:

المشاركة و التعبير عن الرأي بحرية .d1

3. Contents

		25 5 17 120 News	
Teaching	N. Carlotte and the car	o, of lecture	No. of Practical
25.	TOPIC	hours	/ hours /
week			1
1	حقوق الأنسان في القانون الجائي	2	<i>.</i>
•		1	1_00
	المنسان في تغيير جاسته او التخلي عن اجدي	2	
2	خنساته	10	** * 0
The state of the s		at the same	
3	المواثيق الدولية المتعلقة بحماية حقوق الاسياق	2 dil 1	0
J			*·
4	المواثيق الدولية المتعلقة بحماية حقوق الإمسان،	1.2	0
•		and the second second	
	عالفة العوامة والتنمية بالحقوق الاقتصادية و الاجتماعية	2	^
5	و الثقافية	\	0
			-
	المنافقة العوامة في التنبية بالحقوق الاقتصادية و الاجتماعية	$\frac{3}{2}$	0
6	الانتخاب المنتقافية	\[\]	U
		/	
7	المرية المسلم المعالم الدوري الاول	, 2 2	0
			_
8	و المنفوق المنسان و الاجتماعية و الثقافية للانسان	4	0





Course Specifications (2015 – 2016)

9	حقوق الانسان في الشريعة الاسلامية	2	
		L	0
10	حقوق المرأ في قانوني العمل و التأمين الاجتماعي	2	
	A. J. M. A. S. M. C. M. A. S. M. A. S. M. C.		0
11	حقوق الانسان في التقاضي، الحقوق المدنية و السياسية للانسان	2	0
1 Y	الاختبار الدوري الثاني	2	
Total no of			
hours	24		
13	FINALExam		

Teaching and Learning Methods

- 4.1- Lectures (board, data show)
- 4.2- Assignments
- 4.3- Class discussion

Student Assessment Methods

- 5.1. Periodic exams to assess knowledge and understanding as well as intellectual skills.
- 5.2. Written exams to assess all types of skills and mainly general and transferrable skills practice.

Assessment Schedule

Quiz 1

Quiz 2

Final exam

5th week

12th week

12th week; according to semester schedule

Weighting of Assessments

Periodical

Final exam

Total

List of References

6.1- Course Notes: Lecture notes in human rights

5.2- Essential Books (Textbooks):

National human rights strategy the Arab Republic of Egypt







Course Specifications (2015 – 2016)

Facilities required for teaching and learning

- 1. Lecture rooms with data show
- 2. Procurement of latest edition of the above-mentioned texts and others to update the education process

