

## **Polyaniline/banana peel composite: an eco-friendly adsorbent for removal of dihydrogen phosphate from groundwater**

### **Abstract**

A low-cost polyaniline/banana peel (PAni/BP) composite was prepared by oxidative polymerization of aniline in the presence of agricultural waste BP powder and characterized by Fourier-transform infrared spectroscopy, scanning electron microscopy, X-ray diffraction, thermogravimetric analysis, and Brunauer–Emmett–Teller. The surface area and average pore width were  $5.65 \text{ m}^2 \cdot \text{g}^{-1}$  and  $61.2 \text{ nm}$ , respectively. A maximum adsorption capacity of  $56.8 \text{ mg} \cdot \text{g}^{-1}$  and more than 90% removal of  $5.0 \text{ mg} \cdot \text{L}^{-1}$  phosphate, were achieved under optimized conditions. Phosphate adsorption is best described by the pseudo-second-order and Langmuir isotherm models. The Temkin isotherm gave an  $8.2 \text{ J} \cdot \text{mol}^{-1}$  B constant, while the Dubinin–Radushkevich isotherm produced  $14.1 \text{ kJ} \cdot \text{mol}^{-1}$  adsorption energy; both of them supported the chemisorption process. Thermodynamic parameters showed that the phosphate adsorption process was endothermic and spontaneous. For spiked groundwater, removal ranged between 90% and 95%, while desorbed phosphate removal ranged from 83% to 100%, with a relative standard deviation ranging from 2.0% to 7.4%.